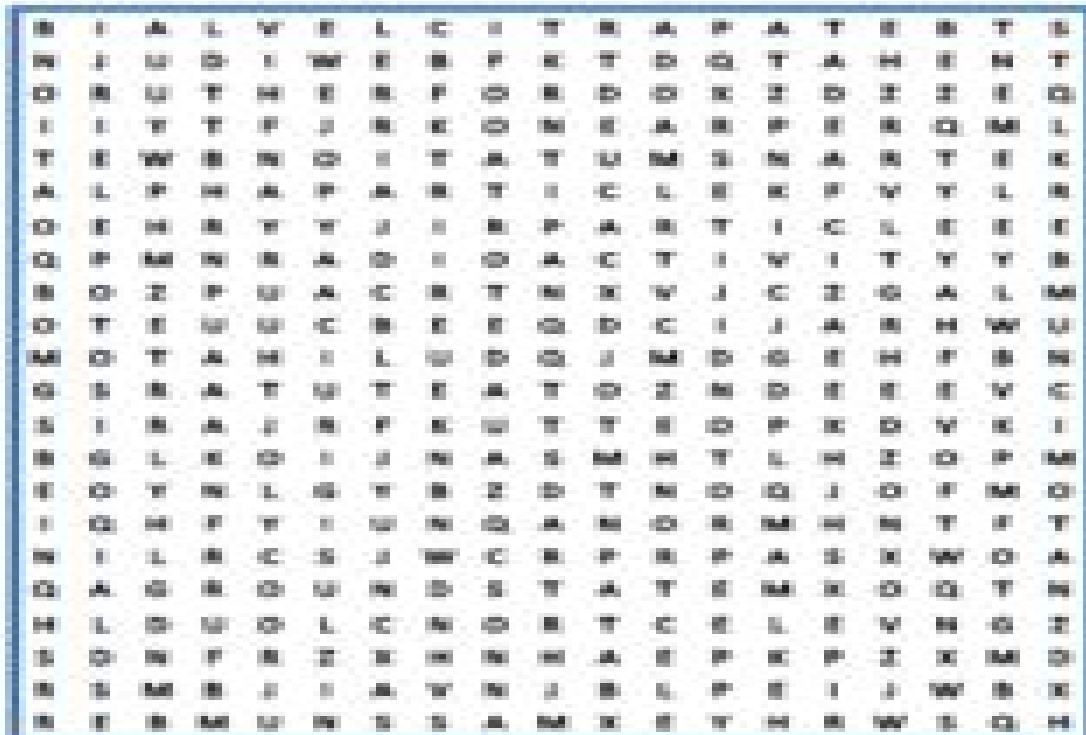


# Atomic Structure

**DIRECTIONS:** Find and circle the vocabulary words and names in the grid. Look for them in all directions including backwards and diagonally.



**ALPHA PARTICLE:** Two neutrons and two protons bound together and produced by radioactive decay.

**ATOM:** The smallest unit of an element.

**ATOMIC NUMBER:** The number of protons in an atom.

**BETA PARTICLE:** Electron emitted by radioactive decay.

**BOHR:** The Danish physicist who made essential contributions to the understanding of atomic structure.

**CHARGE:** How positive or negative an atom is.

**CATHODE RAY:** A beam of electrons.

**DECAY:** To break into smaller pieces.

**ELECTRON:** A negatively charged subatomic particle.

**ELECTRON CLOUD:** Empty space around the nucleus of an atom where electrons travel.

**ELEMENT:** A form of matter that cannot be chemically broken down into simpler substances.

**GROUND STATE:** The lowest possible energy state of an atom.

**HALF-LIFE:** The time for a radioactive substance to lose half of its radioactivity.

**ION:** An atom that has a positive or negative charge due to the gain or loss of one or more electrons.

**ISOTOPE:** The term given to atoms that have the same number of protons but different numbers of neutrons.

**MASS NUMBER:** The total number of protons and neutrons in the nucleus of an atom.

**NEUTRON:** A subatomic particle with no charge.

**NUCLEUS:** The center of an atom.

**ORBITAL:** The region of space around the nucleus where an electron is likely to be found.

**PARTICLE:** A minute quantity of matter.

**PHOTON:** The fundamental particle of visible light.

**PROTON:** A subatomic particle with a positive charge.

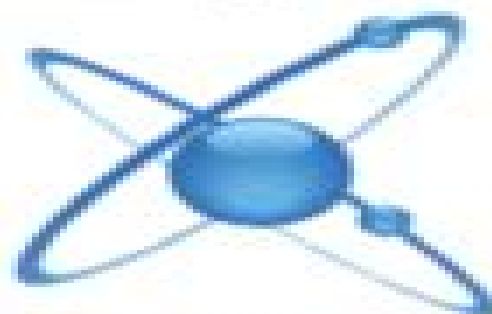
**RADIOACTIVITY:** Emission of particles from an unstable atom.

**RUTHERFORD:** New Zealand physicist who figured out subatomic structure.

**SUBATOMIC:** Smaller than or occurring within an atom.

**THOMSON:** English physicist who discovered the electron.

**TRANSFORMATION:** Changing of one element to another through radioactive decay.



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**R Sanford**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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