

Unit 1 Chemistry Study Guide

History of the Atom:

Democritus (450 BCE):

- Matter is not infinitely divisible.
- There are single particles that cannot be divided further, called **atomos**.

Aristotle (380 BCE):

- Matter is infinitely divisible.
- Matter is called **hyle**.

Antoine Lavoisier (1774):

- **Law of Conservation of Mass:** Mass cannot be created or destroyed.

Joseph Proust (1779):

- **Law of Definite Proportions:** All samples of a pure compound contain the same elements in the same proportion by mass (no matter quantity).

John Dalton (1803 – 1808):

- Observed work of Lavoisier and Proust
- Led Dalton to develop the first **atomic theory**.
- **Dalton's Atomic Theory:**
 - o All matter is composed of extremely small particles called atoms.
 - o All atoms of a given element are identical, having the same size, mass, and chemical properties. The atoms of one element are different from the atoms of all other elements.
 - o Atoms cannot be subdivided, created, or destroyed.
 - o Atoms of different elements can combine in simple whole number ratios to form chemical compounds.
 - o In chemical reactions, atoms are combined, separated or rearranged.
- **Law of Multiple Proportions:** Elements can combine in different proportions. However, the different compounds will have different properties but the ratio of atoms is always a whole number.

Billiard Ball Model:



atom

J.J. Thompson (1906):

- **Cathode Ray Experiment –**
 - o Part 1: Thompson ran electricity through a cathode ray tube and when he put the negatively charged end of a magnet near the tube, he saw the ray of electricity

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Chandresh Agrawal, nandini books



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Teaching Chemistry Around the World Björn Risch, 2010 As teachers we often tend to expect other countries to teach chemistry in much the same way as we do but educational systems differ widely At Bielefeld University we started a project to analyse the approach to chemical education in different countries from all over the world Teaching Chemistry around the World 25 countries have participated in the project The resulting country studies are presented in this book This book may be seen as a contribution to make the structure of chemistry teaching in numerous countries more transparent and to facilitate communication between these countries Especially in the case of the school subject chemistry which is very unpopular on the one hand and occupies an exceptional position on the other hand due to its relevance to jobs and everyday life and most notably due to its importance for innovation capacity and problem solving we have to learn from each others educational systems

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Second International Handbook of Science Education Barry Fraser, Kenneth Tobin, Campbell J. McRobbie, 2011-12-14 The International Handbook of Science Education is a two volume edition pertaining to the most significant issues in science education It is a follow up to the first Handbook published in 1998 which is seen as the most authoritative resource ever produced in science education The chapters in this edition are reviews of research in science education and retain the strong international flavor of the project It covers the diverse theories and methods that have been a foundation for science education and continue to characterize this field Each section contains a lead chapter that provides an overview and synthesis of the field and related chapters that provide a narrower focus on research and current thinking on the key issues in that field Leading researchers from around the world have participated as authors and consultants to produce a resource that is comprehensive detailed and up to date The chapters provide the most recent and advanced thinking in science education making the Handbook again the most authoritative resource in science education

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Resources in Education, 1999-10

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Learning Science Barbara Schneider,Joseph Krajcik,Jari Lavonen,2020-02-11 An innovative internationally developed system to help advance science learning and instruction for high school students This book tells the story of a 3 6 million research project funded by the National Science Foundation aimed at increasing scientific literacy and addressing global concerns of declining science engagement Studying dozens of classrooms across the United States and Finland this international team combines large scale studies with intensive interviews from teachers and students to examine how to transform science education Written for teachers parents policymakers and researchers this book offers solutions for matching science learning and instruction with newly recommended twenty first century standards *Progress of Education in India* ,1898 **Concepts of Matter in Science Education** Georgios Tsaparlis,Hannah Sevan,2013-07-09 Bringing together a wide collection of ideas reviews analyses and new research on particulate and structural concepts of matter Concepts of Matter in Science Education informs practice from pre school through graduate school learning and teaching and aims to inspire progress in science education The expert contributors offer a range of reviews and critical analyses of related literature and in depth analysis of specific issues as well as new research Among the themes covered are learning

progressions for teaching a particle model of matter the mental models of both students and teachers of the particulate nature of matter educational technology chemical reactions and chemical phenomena chemical structure and bonding quantum chemistry and the history and philosophy of science relating to the particulate nature of matter The book will benefit a wide audience including classroom practitioners and student teachers at every educational level teacher educators and researchers in science education If gaining the precise meaning in particulate terms of what is solid what is liquid and that air is a gas were that simple we would not be confronted with another book which while suggesting new approaches to teaching these topics confirms they are still very difficult for students to learn Peter Fensham Emeritus Professor Monash University Adjunct Professor QUT from the foreword to this book **PD, Health and PE** Fay Courtney, David Thomas, 2005

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