

14.3

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate concentrations based on the pH of a given acid or hydrogen ion concentration.
2. Calculate hydrogen ion concentration from pH and hydroxide ion concentration from pOH of a given solution.
3. Calculate the pH of a weak acid solution.

Vocabulary

- | | | |
|-------------------|-------------------------------|---------------------|
| 1. autoionization | 2. hydrogen ion concentration | 3. acid solution |
| 4. autoionization | 5. basic solution | 6. neutral solution |
| | | 7. pH |

Key Equation

- | | |
|-------------------------------------|-------------------------------------|
| 1. $\text{pH} = -\log [\text{H}^+]$ | 2. $\text{pH} = -\log [\text{H}^+]$ |
| 3. $\text{pH} = -\log [\text{H}^+]$ | 4. $\text{pH} = -\log [\text{H}^+]$ |

Part A Completion

Write the hydrogen ion concentration or hydroxide ion concentration of the solutions described. Round the concentration to the correct significant figures. Express values in scientific notation.

1. What substance is _____, contains hydrogen ions?
2. _____ is the concentration of those acids present in stomach acid equal to _____ molar.
3. _____ is the concentration of those acids present in _____ molar _____.
4. _____ is the concentration of those acids present in _____ molar _____.
5. _____ is the concentration of those acids present in _____ molar _____.
6. _____ is the concentration of those acids present in _____ molar _____.
7. _____ is the concentration of those acids present in _____ molar _____.
8. _____ is the concentration of those acids present in _____ molar _____.
9. _____ is the concentration of those acids present in _____ molar _____.
10. _____ is the concentration of those acids present in _____ molar _____.
11. _____ is the concentration of those acids present in _____ molar _____.
12. _____ is the concentration of those acids present in _____ molar _____.
13. _____ is the concentration of those acids present in _____ molar _____.
14. _____ is the concentration of those acids present in _____ molar _____.
15. _____ is the concentration of those acids present in _____ molar _____.
16. _____ is the concentration of those acids present in _____ molar _____.
17. _____ is the concentration of those acids present in _____ molar _____.
18. _____ is the concentration of those acids present in _____ molar _____.
19. _____ is the concentration of those acids present in _____ molar _____.
20. _____ is the concentration of those acids present in _____ molar _____.

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17. _____
18. _____
19. _____
20. _____

Section 19 2 Hydrogen And Ions Acidity Answer Key

Allison Hale, Mary Jo Hovey



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Ophthalmology, E-Book Myron Yanoff, Jay S. Duker, 2022-11-06

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