

## 11

## CHEMICAL REACTIONS

## Practice Problems

*In your notebook, solve the following problems. Use the 3-step problem-solving approach you learned in Chapter 1.*

## SECTION 11.1 DESCRIBING CHEMICAL REACTIONS

1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.
2. Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.
3. Write the skeleton equation representing the heating of magnesium carbonate to produce solid magnesium oxide and carbon dioxide gas.
4. Write a balanced equation for the production of HCl gas from its elements.
5. Write a sentence that completely describes the chemical reaction represented by this balanced equation.



6. Write the word equation for the following equation. Write a sentence fully describing the reaction. Is the equation correctly balanced? Explain.



7. Write a balanced equation representing the formation of aqueous sulfuric acid from water and sulfur trioxide gas.
8. Write a balanced equation from this word equation.  
aqueous silver nitrate + copper metal  $\rightarrow$  silver metal + aqueous copper nitrate
9. Write a balanced equation for the following word equation.



## SECTION 11.2 TYPES OF CHEMICAL REACTIONS

1. Write a balanced equation representing the reaction of magnesium with oxygen gas to produce magnesium oxide.
2. Write the balanced equation for the reaction that occurs between aluminum and fluorine.
3. Write the balanced equation for the production of oxygen gas and potassium chloride from the decomposition of potassium chlorate.
4. Write the balanced equation for the reaction between hydrochloric acid and calcium metal. The products are hydrogen gas and calcium chloride.
5. Write the balanced equation for the combustion of propane ( $\text{C}_3\text{H}_8$ ) to produce carbon dioxide and water vapor.
6. Write the balanced equation for the reaction between iron(III) chloride and sodium hydroxide. The products are iron(III) hydroxide and sodium chloride.

# Section 1describing Chemical Reactions Practice Problems Answers

**Martha Mackin**



## **Section 1describing Chemical Reactions Practice Problems Answers:**

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## **Table of Contents Section 1describing Chemical Reactions Practice Problems Answers**

1. Understanding the eBook Section 1describing Chemical Reactions Practice Problems Answers
  - The Rise of Digital Reading Section 1describing Chemical Reactions Practice Problems Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Section 1describing Chemical Reactions Practice Problems Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Section 1describing Chemical Reactions Practice Problems Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Section 1describing Chemical Reactions Practice Problems Answers
  - Personalized Recommendations
  - Section 1describing Chemical Reactions Practice Problems Answers User Reviews and Ratings
  - Section 1describing Chemical Reactions Practice Problems Answers and Bestseller Lists
5. Accessing Section 1describing Chemical Reactions Practice Problems Answers Free and Paid eBooks
  - Section 1describing Chemical Reactions Practice Problems Answers Public Domain eBooks
  - Section 1describing Chemical Reactions Practice Problems Answers eBook Subscription Services
  - Section 1describing Chemical Reactions Practice Problems Answers Budget-Friendly Options

6. Navigating Section 1describing Chemical Reactions Practice Problems Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Section 1describing Chemical Reactions Practice Problems Answers Compatibility with Devices
  - Section 1describing Chemical Reactions Practice Problems Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Section 1describing Chemical Reactions Practice Problems Answers
  - Highlighting and Note-Taking Section 1describing Chemical Reactions Practice Problems Answers
  - Interactive Elements Section 1describing Chemical Reactions Practice Problems Answers
8. Staying Engaged with Section 1describing Chemical Reactions Practice Problems Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Section 1describing Chemical Reactions Practice Problems Answers
9. Balancing eBooks and Physical Books Section 1describing Chemical Reactions Practice Problems Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Section 1describing Chemical Reactions Practice Problems Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Section 1describing Chemical Reactions Practice Problems Answers
  - Setting Reading Goals Section 1describing Chemical Reactions Practice Problems Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Section 1describing Chemical Reactions Practice Problems Answers
  - Fact-Checking eBook Content of Section 1describing Chemical Reactions Practice Problems Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

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