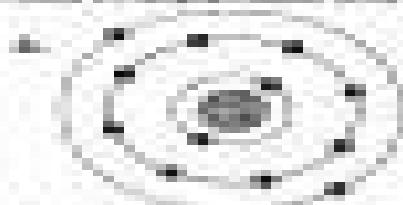


## IPS Unit 9 – Chemical Bonds

### Types of Bonds Worksheet

Directions: Study the diagrams below, then give a brief description of each bond type as shown.



3. Fill in the blanks (use arrows to show the flow).

- Atom (left) electron will move to right? \_\_\_\_\_
- Atom (left) electron will stay in place? \_\_\_\_\_
- What will be the charge of atom left? \_\_\_\_\_
- What will be the charge of atom right? \_\_\_\_\_
- What will be the bond charge of the compound formed? \_\_\_\_\_
- What type of bond will form? \_\_\_\_\_

4. Explain why a covalent bond is stronger than an ionic bond. Why is it stronger?

State and show how the atoms are bonded. (Use arrows to show the flow of electrons.)

Directions: Complete the table comparing the properties of ionic and covalent compounds.

Compounds	Ionic compounds	Covalent compounds
How they compound is formed	Electron transfer from one atom to another	Electron sharing between two atoms
Strongest particle	Crystal	Molecule
How they are held together	Strong	Weak or strong

# Section 2 Reinforcement Types Of Bonds Worksheet

## Answer Key

**Rachel S Tattersall**



## **Section 2 Reinforcement Types Of Bonds Worksheet Answer Key:**

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