

BRILLIANT FIELD SYNERGY ACADEMY
SECOND TERM EXAMINATION 2023/2024 ACADEMIC SESSION

SUBJECT: CHEMISTRY

CLASS: SS1

INSTRUCTION: ANSWER ALL QUESTIONS

1. The following are types of salts EXCEPT (A) acidic (B) basic (C) complex (D) mono salt
2. The gas law which describes the relationship between volume and temperature is (A) Boyle's Law (B) Charles's law (C) Dalton's Law (D) Graham's Law
3. Which of the following phenomena leads to decrease in volume of a liquid in an open container? (A) Brownian motion (B) Diffusion (C) Evaporation (D) Sublimation
4. The following are allotropes of carbon EXCEPT (A) Diamond (B) Graphite (C) Coke (D). Feldspar
5. In which of these substances is the basicity of the acid equal to 1
(A) H_2SO_4 (B) H_3PO_4 (C) CH_3COOH
6. The following are physical properties of acids EXCEPT
(A) acid turns blue litmus paper red (B) concentrated acids are corrosive
(C) acid bleaches substances (D) dilute acids have sour taste
7. Calculate the pressure of Nitrogen gas if its volume is doubled given that its volume is 10dm^3 at a pressure of 6atm (A) 3atm (B) 7atm (C) 5.5atm (D) 3.3atm
8. Which of these shows a mathematical statement of Boyle's law
(A) $PV = \text{constant}$ (B) $P/V = \text{constant}$ (C) $PV^2 = \text{constant}$ (D) $P/T = \text{constant}$
9. A reaction between acids and bases are frequently called (A) combustion
(B) neutralization (C) oxidation (D) substitution
solution (D) petrol
10. Calculate the pH of 0.1mol/dm^3 of solution of HCl (A) 1 (B) 2 (C) 3 (D) 1.5

PART B - ESSAY

INSTRUCTION: Answer only 5 questions

- 1 (i) state Dalton's law of partial pressure (ii) calculate the partial pressures of three gases collected at 760mmHg if the gases are made of 40% oxygen, 30% nitrogen and 30% hydrogen respectively
- 2 (i) state the equation of the general gas law and define all the terms (ii) From the general gas equation, calculate the volume of one mole of an ideal gas at s.t.p ($R = 0.082\text{dm}^3\text{atm K}^{-1}\text{mol}^{-1}$, $T = 273\text{K}$, $p = 1\text{atm}$)

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Lauren Gardner



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Table of Contents Second Term Exam Paper Of Chemistry For Ss1 2015

1. Understanding the eBook Second Term Exam Paper Of Chemistry For Ss1 2015
 - The Rise of Digital Reading Second Term Exam Paper Of Chemistry For Ss1 2015
 - Advantages of eBooks Over Traditional Books
2. Identifying Second Term Exam Paper Of Chemistry For Ss1 2015
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Second Term Exam Paper Of Chemistry For Ss1 2015
 - User-Friendly Interface
4. Exploring eBook Recommendations from Second Term Exam Paper Of Chemistry For Ss1 2015
 - Personalized Recommendations

- Second Term Exam Paper Of Chemistry For Ss1 2015 User Reviews and Ratings
- Second Term Exam Paper Of Chemistry For Ss1 2015 and Bestseller Lists
- 5. Accessing Second Term Exam Paper Of Chemistry For Ss1 2015 Free and Paid eBooks
 - Second Term Exam Paper Of Chemistry For Ss1 2015 Public Domain eBooks
 - Second Term Exam Paper Of Chemistry For Ss1 2015 eBook Subscription Services
 - Second Term Exam Paper Of Chemistry For Ss1 2015 Budget-Friendly Options
- 6. Navigating Second Term Exam Paper Of Chemistry For Ss1 2015 eBook Formats
 - ePub, PDF, MOBI, and More
 - Second Term Exam Paper Of Chemistry For Ss1 2015 Compatibility with Devices
 - Second Term Exam Paper Of Chemistry For Ss1 2015 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Second Term Exam Paper Of Chemistry For Ss1 2015
 - Highlighting and Note-Taking Second Term Exam Paper Of Chemistry For Ss1 2015
 - Interactive Elements Second Term Exam Paper Of Chemistry For Ss1 2015
- 8. Staying Engaged with Second Term Exam Paper Of Chemistry For Ss1 2015
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Second Term Exam Paper Of Chemistry For Ss1 2015
- 9. Balancing eBooks and Physical Books Second Term Exam Paper Of Chemistry For Ss1 2015
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Second Term Exam Paper Of Chemistry For Ss1 2015
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Second Term Exam Paper Of Chemistry For Ss1 2015
 - Setting Reading Goals Second Term Exam Paper Of Chemistry For Ss1 2015
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Second Term Exam Paper Of Chemistry For Ss1 2015
 - Fact-Checking eBook Content of Second Term Exam Paper Of Chemistry For Ss1 2015

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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