

Chapter 11: Gas Law Worksheet Answer Key

Complete the following calculation by list the given information, rewriting the formula to solve for the unknown, and plugging in the information (number and units), and writing the answer significantly.

1. A sample of 0.500 moles of gas is placed in a container of volume of 2.50 L. What is the pressure of the gas in torr if the gas is at 25 °C? (Ideal Gas Law)

$$P = \frac{nRT}{V} = \frac{(0.500 \text{ moles}) (0.0821 \text{ L} \cdot \text{atm}) (298 \text{ K})}{(\text{mole} \cdot \text{K}) (2.50 \text{ L})} = P = 4.89 \text{ atm} = 3720 \text{ torr}$$

2. If 2.00 mol of gas occupies 4.50 L at STP. How much of the same gas will occupy 3.00 L at STP? (Avogadro's Law)

$$n_2 = \frac{V_2 n_1}{V_1} = \frac{(3.00 \text{ L}) (2.00 \text{ moles})}{(4.50 \text{ L})} = 1.33 \text{ moles}$$

3. Determine the partial pressures of each of the gases in the following mixture: 17.04 g NH₃, 40.36 g Ne and 19.00 g F₂. The gases are at 1.50 atm of pressure. (Dalton's Law of Partial Pressure & Mole Ratios)

$$\frac{17.04 \text{ g NH}_3}{17.04 \text{ g NH}_3} \left| \frac{1 \text{ mole NH}_3}{17.04 \text{ g NH}_3} \right| \left| \frac{1.000 \text{ mole NH}_3}{3.500 \text{ moles}} \right| \quad \frac{1.000 \text{ mole NH}_3}{3.500 \text{ moles}} = 0.2857 \times 1.50 \text{ atm} = 0.429 \text{ atm NH}_3$$

$$\frac{40.36 \text{ g Ne}}{20.18 \text{ g Ne}} \left| \frac{1 \text{ mole Ne}}{20.18 \text{ g Ne}} \right| \left| \frac{2.000 \text{ moles Ne}}{3.500 \text{ moles}} \right| \quad \frac{2.000 \text{ moles Ne}}{3.500 \text{ moles}} = 0.5714 \times 1.50 \text{ atm} = 0.857 \text{ atm Ne}$$

$$\frac{19.00 \text{ g F}_2}{38.00 \text{ g F}_2} \left| \frac{1 \text{ mole F}_2}{38.00 \text{ g F}_2} \right| \left| \frac{0.5000 \text{ moles F}_2}{3.500 \text{ moles}} \right| \quad \frac{0.5000 \text{ moles F}_2}{3.500 \text{ moles}} = 0.1429 \times 1.50 \text{ atm} = 0.214 \text{ atm F}_2$$

$$1.000 \text{ moles} + 2.000 \text{ moles} + 0.5000 \text{ moles} = 3.500 \text{ moles}$$

4. If at 1.00 atm of pressure water boils at 100. °C, at what temperature would water boil if the pressure is 600. torr? (This shows why food doesn't cook well at higher elevations) (Gay-Lussac's Law)

$$T_2 = \frac{T_1 P_2}{P_1} = \frac{(373 \text{ K}) (0.789 \text{ atm})}{(1.00 \text{ atm})} = 294 \text{ K}$$

5. Calculate the volume of 40.6 g of F₂ at STP. (Ideal Gas Law)

$$\frac{40.6 \text{ g F}_2}{38.00 \text{ g F}_2} \left| \frac{1 \text{ mole F}_2}{38.00 \text{ g F}_2} \right| \left| \frac{22.4 \text{ L F}_2}{1 \text{ mole F}_2} \right| \quad \frac{22.4 \text{ L F}_2}{1 \text{ mole F}_2} = 23.9 \text{ L F}_2$$

or

$$\frac{40.6 \text{ g F}_2}{38.00 \text{ g F}_2} \left| \frac{1 \text{ mole F}_2}{38.00 \text{ g F}_2} \right| \left| \frac{1.07 \text{ mole F}_2}{1.07 \text{ mole F}_2} \right| \quad \frac{1.07 \text{ mole F}_2}{1.07 \text{ mole F}_2} = 24.0 \text{ L F}_2$$

$$V = \frac{nRT}{P} = \frac{(1.07 \text{ moles}) (0.0821 \text{ L} \cdot \text{atm}) (273 \text{ K})}{(\text{mole} \cdot \text{K}) (1.00 \text{ atm})} = 24.0 \text{ L F}_2$$

6. If the winter pressure that most car tire tires should be at to wear evenly is 32.0 psi, what is the pressure in atm? (Pressure Conversion)

$$\frac{32.0 \text{ psi}}{14.7 \text{ psi}} \left| \frac{1.00 \text{ atm}}{14.7 \text{ psi}} \right| \quad \frac{1.00 \text{ atm}}{14.7 \text{ psi}} = 2.18 \text{ atm}$$

Section 13 Gases Worksheet Answer Key

Vicki L. Hackett, Paul C. Dalmas

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the authors' names.

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Publishing,2010-09-01 Designed to work with both differentiated levels of Think Green these 24 page guides were developed in consultation with several state educational standards and contain multiple components Three lesson plans are included These lesson plans are divided into sections vocabulary preview reading the text discussing the meaning word work extending the meaning and critiquing The teacher s guide also includes 11 worksheets 2 vocabulary 1 writing 1 index 2 review and 5 activity sheets *Teacher's Wraparound Edition: Twe Biology Everyday Experience* Albert Kaskel,1994-04-19

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Health Unit Coordinating Elaine A. Gillingham,Monica Wadsworth Seibel,2013-01-24 Get the most comprehensive in depth coverage on health unit coordinating from the industry's most popular text Expert authors Elaine Gillingham and Monica

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Official Gazette of the United States Patent Office United States. Patent Office,1957 **Operating Section Proceedings** American Gas Association. Operating Section,1981 **Bold Ventures** S. Raizen,E.D. Britton,1996-12-31 This book presents comprehensive results from case studies of five innovations in science education that have much to offer toward understanding current reforms in this field Each chapter tells the story of a case in rich detail with extensive documentation and in the voices of many of the participants the innovators the teachers the students Similarly Volume 3 of **Bold Ventures** pre sents the results from case studies of five innovations in mathematics education Volume 1 provides a cross case analysis of all eight innovations Many U S readers certainly will be very familiar with the name of at least one if not all of the science innovations discussed in this volume for example Project 206l and

probably with their general substance Much of the education community s familiarity with these arises from the projects own dissemination efforts The research reported in this volume however is one of the few detailed studies of these innovations undertaken by researchers outside the projects them selves Each of the five studies was a large scale effort involving teams of researchers over three years These teams analyzed many documents attended numerous critical project meetings visited multiple sites conducted dozens of individual interviews The team leaders Atkin Huberman Rowe having spent much time with science education over long careers looked at these innovations through many lenses It was a daunting task for each team to sift through the mountains of detail in order to bring the most compelling themes to the surface *Statistics of Income* United States. Internal Revenue Service,1992

The Top Books of the Year Section 13 Gases Worksheet Answer Key The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous captivating novels enthralling the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the fascinating narratives that have charmed audiences this year. Section 13 Gases Worksheet Answer Key : Colleen Hoover "It Ends with Us" This touching tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Section 13 Gases Worksheet Answer Key : Delia Owens "Where the Crawdads Sing" This mesmerizing coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a exceptional and gripping novel that will keep you wondering until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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