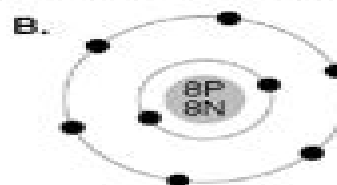
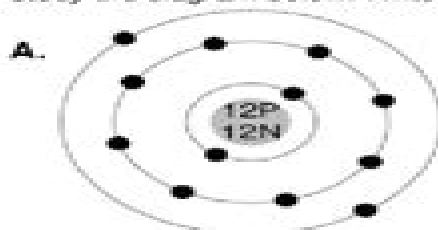


Name: _____ Period: _____

IPS Unit 9 – Chemical Bonds

Types of Bonds Worksheet

Directions: Study the diagram below. Write your answers to the questions in the spaces provided.



1. If atom A loses electrons to atom B.
 - a. how many electrons will atom A lose? _____
 - b. how many electrons will atom B gain? _____
 - c. what will be the charge of atom A? _____
 - d. what will be the charge of atom B? _____
 - e. what will be the total charge of the compound formed?

 - f. what type of bond will form?

2. Explain why an element's charge is related to the group on the periodic table to which it belongs.

Directions: Complete the table comparing ionic compounds and covalent compounds.

Characteristic	Ionic compounds	Covalent compounds
How the compound is formed		
Smallest particle		
Usual state at room temperature		

Section 2 Types Of Bonding Chapter 20 Answers

**DM Vasudevan, Sreekumari S, Kannan
Vaidyanathan**



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Survey and Study of Administrative Organization, Procedure, and Practice in the Federal Agencies by the Committee on Government Operations United States. Congress. House. Committee on Government Operations, 1957

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