

Assessment

Chapter Test B

Chemistry of Life

Write the correct letter in the blank before each numbered term.

- | | |
|------------------------|--|
| _____ 1. covalent bond | a. involves transferred electrons |
| _____ 2. atom | b. loss of an electron |
| _____ 3. reduction | c. substance that cannot be broken down |
| _____ 4. compound | d. involves shared electrons |
| _____ 5. molecule | e. a substance composed of two or more elements |
| _____ 6. ionic bond | f. gain of an electron |
| _____ 7. oxidation | g. simplest part of an element |
| _____ 8. element | h. simplest part of a substance that can exist in a free state |

Write the letter of the most correct answer in the blank.

- _____ 9. Elements tend to undergo chemical reactions that _____ their stability.
- | | |
|-------------|------------------|
| a. increase | c. do not change |
| b. decrease | d. disrupt |
- _____ 10. Sodium chloride (table salt) is an example of a compound formed by _____.
- | | |
|----------------------|-----------------------------|
| a. hydrogen bonding. | c. ionic bonding. |
| b. covalent bonding. | d. electromagnetic bonding. |
- _____ 11. When pure water dissociates, it forms equal numbers of _____.
- | |
|---------------------------------------|
| a. hydroxide ions and sodium ions. |
| b. hydronium ions and water ions. |
| c. hydroxide ions and water ions. |
| d. hydroxide ions and hydronium ions. |
- _____ 12. Matter can change from a solid to a liquid by the addition of energy, which makes the particles in the matter _____.
- | | |
|-----------------|----------------------------------|
| a. move faster. | c. dissociate. |
| b. move slower. | d. increase their specific heat. |
- _____ 13. Charged particles that move around an atom's nucleus are _____.
- | | |
|---------------|--------------|
| a. electrons. | c. ions. |
| b. protons. | d. neutrons. |

Section 21 2 Review Modern Biology Answers

Michael Brown



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