

15. Describe the two diagrams of a bottled carbonated beverage below as *greater pressure* or *lower pressure*, and then as *greater solubility* or *lower solubility*. How do these two examples illustrate the relationship between the solubility of a gas and its vapor pressure?

In the closed bottle, the pressure is high and lots of  $\text{CO}_2$  is dissolved in the  
carbonated beverage. Once the cap is removed, the vapor pressure decreases,  
allowing more  $\text{CO}_2$  in the carbonated beverage to escape.



greater pressure and greater solubility      lower pressure and lower solubility

16. How does a solution become supersaturated? If the temperature of a solution having  
a small excess of solid solute is raised, the solute dissolves. If the solution is then  
allowed to cool slowly, the excess solute may stay dissolved at a temperature below  
the temperature at which it would ordinarily crystallize.

## SECTION 16.2 CONCENTRATIONS OF SOLUTIONS (pages 480–486)

*This section explains how to solve problems involving molarity of a solution, how to prepare dilute solutions from more concentrated solutions, and what is meant by percent by volume and percent by mass.*

### ► Molarity (pages 480–482)

1. A measure of the amount of solute dissolved in a given quantity of solvent is the concentration of a solution.
2. The most important unit of concentration in chemistry is molarity.
3. Is the following sentence true or false? Molarity is the number of moles of dissolved solute per liter of solvent. false

## Section 16 3 Colligative Properties Of Solutions Key

**SA Adler**



## **Section 16 3 Colligative Properties Of Solutions Key:**

## Unveiling the Power of Verbal Beauty: An Psychological Sojourn through **Section 16 3 Colligative Properties Of Solutions Key**

In a global inundated with monitors and the cacophony of immediate connection, the profound energy and psychological resonance of verbal artistry frequently fade in to obscurity, eclipsed by the continuous assault of noise and distractions. Yet, nestled within the lyrical pages of **Section 16 3 Colligative Properties Of Solutions Key**, a interesting work of literary brilliance that impulses with organic emotions, lies an remarkable journey waiting to be embarked upon. Written by a virtuoso wordsmith, this exciting opus manuals readers on a mental odyssey, lightly exposing the latent possible and profound impact embedded within the delicate internet of language. Within the heart-wrenching expanse of this evocative evaluation, we will embark upon an introspective exploration of the book is key styles, dissect their captivating publishing design, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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