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Guided Practice - Acids & Bases

Properties of Acids and Bases

- 1. Match the following terms with their descriptions:
 - is. Arrhenius Acid ______ the substance formed when a proton is added to
 - a base
 - Arrhenius Base ______ the remaining substance when a proton is lost
 - from an acid
 - Conjugate Acid ______ a substance that produces OH in solution
 - d. Conjugate Base a substance producing H* in solution
 - e. Bonsted-Lowry Acid ______ a proton acceptor
 - f. Bronsted-Lowry Base a proton donor
- Which of the following represent conjugate acid-base pairs? Label each species as a conjugate acid or a conjugate base. For those pairs that are NOT conjugates, write the correct conjugate acid or base for each species in the pair, liabeling each species correctly.
 - a HSON SON
 - b. Hibr. Bro-
 - e. Hypout Pour
 - d. H890's 190"
- 3. Complete the following table regarding acid strength.

Property	Strong Acid	Wesk Acid
Substation		
Describe Dissociation		
Species present in solution		
Relative strength of corpugate base compared to mater		
Equation showing acid's reaction to mater		
Paragraphy List in acids		

Section 19 Acids And Bases Answer Key

Arshad Iqbal

Section 19 Acids And Bases Answer Key:

Chemistry Therald Moeller,2012-12-02 Chemistry with Inorganic Qualitative Analysis is a textbook that describes the application of the principles of equilibrium represented in qualitative analysis and the properties of ions arising from the reactions of the analysis This book reviews the chemistry of inorganic substances as the science of matter the units of measure used atoms atomic structure thermochemistry nuclear chemistry molecules and ions in action This text also describes the chemical bonds the representative elements the changes of state water and the hydrosphere which also covers water pollution and water purification Water purification occurs in nature through the usual water cycle and by the action of microorganisms The air flushes dissolved gases and volatile pollutants when water seeps through the soil it filters solids as they settle in the bottom of placid lakes Microorganisms break down large organic molecules containing mostly carbon hydrogen nitrogen oxygen sulfur or phosphorus into harmless molecules and ions This text notes that natural purification occurs if the level of contaminants is not so excessive This textbook is suitable for both chemistry teachers and students

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