Chapter 4

Class_

Date

Chapter 4 Atomic Structure

Section 4.1 Studying Atoms (pages 100-105)

This section discusses the development of atomic models.

Reading Strategy (page 100)

Summarizing As you read, complete the table about atomic models. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.

Atomic Models			
Scientist	Evidence	Model	
72	Patio of masses in compounds		
	Deflected beam		
Rutherford		Positive, dense nucleus	

Ancient Greek Models of Atoms (page 100)

- Democritus named the smallest particles of matter _______ because they could not be divided.
- 2. List the four elements that Aristotle included in his model of matter.

a. ______ b.

d. ____

Dalton's Atomic Theory (page 101)

- Is the following sentence true or false? John Dalton gathered evidence for the existence of atoms by measuring the masses of elements that reacted to form compounds.
- 4. What theory did Dalton propose to explain why the elements in a compound always join in the same way?
- Circle the letters of the sentences that represent the main points of Dalton's theory of atoms.
 - a. All elements are composed of atoms.
 - In a particular compound, atoms of different elements always combine the same way.
 - c. All atoms have the same mass.
 - d. Compounds contain atoms of more than one element.

Section 1 Studying Atoms Answers

JL Elias

Section 1 Studying Atoms Answers:

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