In the list below, circle the properties of acids and box the properties of bases.

Sour taste Feel slippery Contains hydroxide ions

Turns litmus blue Reacts with metals to form Ho Turns litmus red

Bitter taste Contains hydrogen ions Corrosive

Complete the chart below:

Acid-Base Definitions		
Type	Acid Definition	Base Definition
Arrhenius	Donates an H' ion in solution	Donates an OH ion in solution
Bronsted-Lowry	Donates an H* ion	Accepts an H* ion
Lewis	Accepts an electron pair	Donates an electron pair

- What term(s) describe a substance that can act as an acid or a base? Amphoteric or amphiprotic
- 4. Neutralization reactions always produce a SALT and WATER
- If the [H^{*}] is greater than the [OH^{*}], the solution is acidic basic neutral
- 6. If the [OH] is greater than the [H], the solution is acidic basic neutral
- 7. If the [H'] is equal to the [OH'], the solution is acidic basic neutral
- 8. If the pH is 7, the solution is acidic basic neutral
- If the pH is greater than 7, the solution is acidic basic neutral
- 10. If the pH is less than 7, the solution is acidic basic neutral
- 11. How many moles of lithium hydroxide are required to neutralize 19.2 grams of chloric acid?

(19.2 g HClO₂) x (1 mol HClO₂/84 g HClO₂) x (1 mol LiOH/1 mol HClO₂) = 0.228 mol LiOH

12. How many grams of calcium hydroxide are required to neutralize 132.4 grams of nitric acid?

 $Ca(OH)_2 + 2 HNO_3 \rightarrow Ca(NO_3)_2 + 2 H_2O$

(132.4 g HNO₃) x (1 mol HNO₃/63 g HNO₃) x (1 mol Ca(OH)₂/2 mol HNO₃) x (74 g Ca(OH)₂/1 mol Ca(OH)₂) = <u>77.76 g Ca(OH)₂</u>

Section 20 Describing Acids And Bases Answers

Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag

Section 20 Describing Acids And Bases Answers:

AP Chemistry For Dummies Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out or your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP Chemistry For Dummies to build your confidence and test taking know how so you can ace that exam Physical Chemistry and Its Biological Applications Wallace Brey, 2012-12-02 Physical Chemistry and Its Biological Applications presents the basic principles of physical chemistry and shows how the methods of physical chemistry are being applied to increase understanding of living systems Chapters 1 and 2 of the book discuss states of matter and solutions of nonelectrolytes Chapters 3 to 5 examine laws in thermodynamics and solutions of electrolytes Chapters 6 to 8 look at acid base equilibria and the link between electromagnetic radiation and the structure of atoms Chapters 9 to 11 cover different types of bonding the rates of chemical reactions and the process of adsorption Chapters 12 to 14 present molecular aggregates magnetic resonance spectroscopy and photochemistry and radiation This book is useful to biological scientists for self study and reference With modest additions of mathematical material by the teacher the book should also be suitable for a full year major s course in physical chemistry Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection

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instrumental analysis aka Analytical Chemistry The goal of the text is to provide a foundation of the analytical process tools and computational methods and resources and to illustrate with problems that bring realism to the practice and importance of analytical chemistry It is designed for undergraduate college students majoring in chemistry and in fields related to chemistry *Chemistry* John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant *Tariff Information*, 1921 United States. Congress. House. Committee on Ways and Means, 1921

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