

Matter Unit Study Guide [1]

Name _____ Date _____ Period _____

1. Oxygen and Hydrogen combine to form water. In the space to the right, draw an illustration of a water molecule. Additionally, identify water as any of the following that apply: an atom, an element, a molecule, and/or a compound. Explain your answer. SGP1a

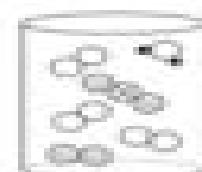
2. What is the smallest particle of the element from (P) that can still be classified as intact? SGP1a

3. Paper, glass, and iron are all made up of _____ SGP1a

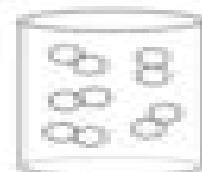
4. A molecule is to a compound as an atom is to an _____ SGP1a

5. The diagram shows three containers of gas. Use the diagram to answer the questions below. SGP1a-f

- a. Which container(s) has atoms?



Container R



Container S



Container T

- b. Which container(s) has one element?

Key
● Type A
● Type B
● Type C
■ Type D

- c. Which container(s) has a pure substance?

- d. Which container(s) has molecules?

- e. Which container(s) has a mixture?

- f. Which container(s) has components?

6. Which of the following does not represent a compound: CH₄, NaCl, CO₂, H₂, NaOH
Explain your answer. SGP1b

7. A solution of saltwater sits in the sun for 6 days. After 6 days, only salt remains in the cup. Explain why this is an example of a mixture. SGP1b

8. Define 'Water' by writing all the following: element, compound, mixture. Explain your answer. SGP1b

9. Which of the following does not belong: Air, Salt, Water, or Oxygen. Explain. SGP1b

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