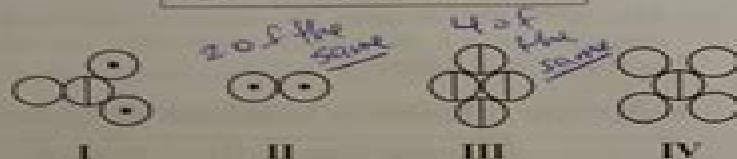
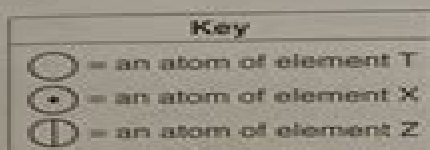


39 Given four particle models:



Which two models can be classified as elements?

- (1) I and II  
 (2) I and IV  
 (3) II and III  
 (4) II and IV

40 After being thoroughly stirred at  $10.^{\circ}\text{C}$ , which mixture is heterogeneous?

- (1) 25.0 g of KCl and 100. g of  $\text{H}_2\text{O}$  *unsat.*  
 (2) 25.0 g of  $\text{KNO}_3$  and 100. g of  $\text{H}_2\text{O}$  *supersat.*  
 (3) 25.0 g of NaCl and 100. g of  $\text{H}_2\text{O}$  *unsat.*  
 (4) 25.0 g of  $\text{NaNO}_3$  and 100. g of  $\text{H}_2\text{O}$  *unsat.*

41 Which two compounds are electrolytes?

- (1) KOH and  $\text{CH}_3\text{COOH}$   
 (2) KOH and  $\text{C}_3\text{H}_{12}$   
 (3)  $\text{CH}_3\text{OH}$  and  $\text{CH}_3\text{COOH}$   
 (4)  $\text{CH}_3\text{OH}$  and  $\text{C}_3\text{H}_{12}$

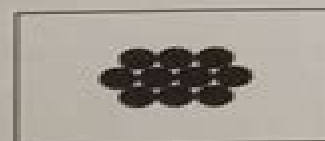
42 Which statement explains why a  $\text{CO}_2$  molecule is nonpolar?

- (1) Carbon and oxygen are both nonmetals.  
 (2) Carbon and oxygen have different electronegativities.  
 (3) The molecule has a symmetrical distribution of charge.  
 (4) The molecule has an asymmetrical distribution of charge.

43 Which temperature change indicates an increase in the average kinetic energy of the molecules in a sample?

- (1)  $15.^{\circ}\text{C}$  to  $298\text{ K}$   $\uparrow$   
 (2)  $37.^{\circ}\text{C}$  to  $273\text{ K}$   $\downarrow$   
 (3)  $305\text{ K}$  to  $0.^{\circ}\text{C}$   $\downarrow$   
 (4)  $355\text{ K}$  to  $25.^{\circ}\text{C}$   $\downarrow$

44 Given the particle diagram:



Which substance at STP can be represented by this particle diagram? *Solid*

- (1)  $\text{N}_2$  *(g)*  
 (2)  $\text{H}_2$  *(g)*  
 (3)  $\text{Mg}$  *(s)*  
 (4)  $\text{Kr}$  *(g)*

45 Which type of equilibrium exists in a sealed flask containing  $\text{Br}_2(\ell)$  and  $\text{Br}_2(\text{g})$  at  $298\text{ K}$  and  $1.0\text{ atm}$ ?

- (1) static phase equilibrium  
 (2) static solution equilibrium  
 (3) dynamic phase equilibrium  
 (4) dynamic solution equilibrium

46 What are the products when potassium hydroxide reacts with hydrochloric acid?

- (1)  $\text{KH}(\text{s})$ ,  $\text{Cl}^+(\text{aq})$ , and  $\text{OH}^-(\text{aq})$   
 (2)  $\text{K}(\text{s})$ ,  $\text{Cl}_2(\text{g})$ , and  $\text{H}_2\text{O}(\ell)$   
 (3)  $\text{KCl}(\text{aq})$  and  $\text{H}_2\text{O}(\ell)$   
 (4)  $\text{KOH}(\text{aq})$  and  $\text{Cl}_2(\text{g})$

47 In a titration, 20.0 milliliters of a  $0.150\text{ M}$   $\text{NaOH}(\text{aq})$  solution exactly neutralizes 24.0 milliliters of an  $\text{HCl}(\text{aq})$  solution. What is the concentration of the  $\text{HCl}(\text{aq})$  solution?

- (1)  $0.125\text{ M}$   
 (2)  $0.180\text{ M}$   
 (3)  $0.250\text{ M}$   
 (4)  $0.360\text{ M}$

48 What fraction of a  $\text{Sr-90}$  sample remains unchanged after 87.3 years?

- (1)  $\frac{1}{2}$   
 (2)  $\frac{1}{3}$   
 (3)  $\frac{1}{4}$   
 (4)  $\frac{1}{8}$

*Handwritten calculation:*  

$$3 = \frac{87.3}{29.14}$$

$$\frac{1}{2} = \frac{1}{8}$$



# Unit 6 Answer For Chemistry Regents Review

**Barron's Educational Series, Gregory  
Scott Hunter**





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**Review** G. Scott Hunter, Scott Hunter, 2001 A high school biology supplement that meets New York State's curriculum standards containing concept reviews with question sets and explained answers an overview of required laboratory skills a glossary and two sample Living Environment Regents Exams with answers *New York Times Saturday Review of Books and Art*, 1970      **Chemistry and Industry Review**, 1958      New York State Education, 1937



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web competency statement ii to advance the physical and intellectual competence functional area 4 physical in my program i provide opportunities for children to develop

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web competency statement 2 directions this reflective statement must relate directly to the competency standard ii to advance physical and intellectual competence the

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web competency statement i self as a preschool teacher i support my children by helping them to identify their own feelings i encourage and acknowledge every child throughout

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web the candidate is required to write a reflective statement of competence for each of the competency goals a total of six 6 each reflective statement of competence

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web cda competency 1 to promote good health practices i check each child entering care for symptoms of illness or injury everyone washes hands before entering to play and often

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web take the time to write them well your statements are intended to reflect your own teaching practices in each of the cda competency standards areas reflective statements of

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web competency statement vi by cda class participant to maintain a commitment to professionalism i am going to maintain a commitment to professionalism by pledging to

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web each reflective competency statement should start with an preamble paragraph these paragraph needs to include what the goal is and wherewith you meet it is the courses of

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web competency statement v by cda class participant to ensure a well run purposeful program responsive to participant needs i believe that in order to ensure a well run

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web feb 12 2017 a competency statement 1 for a cda should include a description of the candidate s knowledge skills and abilities related to the cda credential it should also

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web cda competency statement i directions prepare a reflective statement about how you establish and maintain a safe healthy learning environment this statement should be

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web competency statement i to establish a safe healthy learning environment safe to maintain a safe environment i will ensure that my classroom always has the proper adult

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