

Sharada English High School, Marcela Goa  
UNIT TEST  
(PHYSICAL AND CHEMICAL CHANGES)

Q.4. Complete the following sentences by choosing the correct options given below:

(1/2 mark each =  $\frac{1}{2} \times 4 = 2$  marks)

1. Melting of wax is an example of Physical change  
a. Physical change  
b. Physical and Chemical change  
c. Chemical change  
d. chemical property

$\frac{1}{2}$

2. When Magnesium ribbon is burnt in presence of oxygen Magnesium oxide powder is formed.  
a. Magnesium Hydroxide  
b. Calcium oxide  
c. Calcium hydroxide  
d. Magnesium oxide

$\frac{1}{2}$

3. The Copper Sulphate solution turns green in colour due to the formation of copper sulphate  
a. Iron Sulphate  
b. Copper Sulphate  
c. Iron hydroxide  
d. Copper hydroxide

$\frac{1}{2}$

4. In rusting of iron, rust is formed which is a substance called as Iron oxide  
a. Iron nitrate  
b. Calcium carbonate  
c. Iron oxide  
d. Calcium hydroxide

Q.5. Give Reasons:

(2 marks each =  $2 \times 2 = 4$  marks)

1. The rust formation is faster on ships inspite of being painted.

Because salt water reacts with iron that is faster than form rust. This is why the rust formation is faster on ships inspite of being painted.

2

2. Stainless utensils made up of steel are free from rust.

Chromium Because inside the stainless steel made up of zinc and chromium the zinc and chromium is a property of iron in which they are free of rust and mixing of iron.

1

THANK YOU



# Unit Chapter Test

**Ashwin Pajankar**



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