Name:	Dute: Pd:
	Chemical Reactions CTICE TEST/ REVIEW
I. A chemical reaction is a process in wh	nich
a. products change into reactants	c. substances can change state
b. the law of conservation of mass applies	d. all of these
2. During a chemical reaction,	
a. new elements are produced	c. atoms are destroyed
b. atoms are rearranged	d. elements are destroyed
3. An equation is balanced by	
a. changing subscripts	c. erasing elements as necessary
b. adding coefficients	d. adding elements as necessary
4. What are the reactants in the following	g chemical equation:
Zn + CuSO+ → ZnSO++ Cu	
a. zinc and copper	c. zinc and copper (II) sulfate
b. zinc sulfate and copper	d. only zinc
5. What are the products in the above eq	uation?
a. zinc and copper	c. zinc and copper (II) sulfate
b. zinc sulfate and copper	d. only zinc
6. The breaking down of a substance into	o two or simpler substances is
a. Displacement b. Synthesis	c. Decomposition d. Double displacement
7. Each substance to the right of the arro	w in a chemical equation is a(n)
a. Reactant	c. Product
b. Catalyst	d. Inhibitor
8. According to the law of conservation of many atoms of hydrogen must be part of the present of the prese	of mass, if two atoms of hydrogen are used as a reactant, how reduct?
a. 4 b. 2	c. 1 d. 0
	following chemical equation, NH ₂ Cl → NH ₃ + HCl?
a. Displacement b. Synthesis	c. Decomposition d. Double displacement

Unit 7 Chemical Reactions Test

Deepa Chaudhary,Virendra,Priyanka Sharm,Rashmi Sharma,Ashwini Kumar Prasad

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