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Chapter 7

Equilibria for Multiple Component Systems

- 7.2. With only a single component and phase, if only $\text{Fe}_2(\text{SO}_4)_3$ were present in our system, we would have $C - P + 2 = 1 - 1 + 2 = 2$ degrees of freedom.
- 7.4. For a single-component system, there will never be negative degrees of freedom. The maximum number of phases that can exist in equilibrium is three, so the Gibbs phase rule would yield $1 - 3 + 2 = 0$ degrees of freedom (like might occur at the triple point). For a two-component system, the maximum number of phases that can exist in equilibrium is four. There will never be negative degrees of freedom.
- 7.6. In this reaction, there are only three independent components (the amount of the fourth can be determined from the equilibrium constant) and 3 phases, two solids and a gas mixture. Therefore, there are $3 - 3 + 2 = 2$ degrees of freedom.
- 7.8. For water: $n = \frac{pV}{RT} = \frac{(23.76 \text{ torr})(5.00 \text{ L})}{(0.08205 \text{ L} \cdot \text{atm/mol} \cdot \text{K})(298.15 \text{ K})} \times \frac{1 \text{ atm}}{760 \text{ torr}} = 6.39 \times 10^{-3} \text{ mol}$ of water is needed to ensure that there is a liquid and gas phase. This equals 0.115 grams.
For methanol:
- $$n = \frac{pV}{RT} = \frac{(125.0 \text{ torr})(5.00 \text{ L})}{(0.08205 \text{ L} \cdot \text{atm/mol} \cdot \text{K})(298.15 \text{ K})} \times \frac{1 \text{ atm}}{760 \text{ torr}} = 3.36 \times 10^{-2} \text{ mol}$$
- of methanol is needed to ensure that there is a liquid and gas phase. This equals 1.08 grams.
- 7.10. Since, according to equation 7.11, $a_i = \frac{p_i}{p_i^*}$, and since $p_i^* = 760 \text{ torr}$, we can calculate a_i :

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Ruth W. Chabay, Bruce A. Sherwood



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Methods of Analysis Second Edition continues to be the most current detailed comprehensive tool for researchers and practitioners working with soil Fundamentals of Physics, Chapters 33-37 David Halliday, Robert Resnick, Jearl Walker, 2010-03

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web the practice of occupational hygiene the classical steps in occupational hygiene practice are the recognition of the possible health hazards in the work environment the evaluation of hazards which is the process of assessing exposure and reaching conclusions as to the level of risk to human health

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web to measure worker exposures assess worker risks and develop controls to improve the workplace environment occupational hygienists can provide advice on control strategies including hazard elimination engineering modifications administrative controls and finally personal protective equipment

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web occupational hygiene is a set of systems used to determine the acceptable hazard limit in the working environment and to maintain a healthy life in the worker acceptable level is expressed as the amount that is believed to be safe in the defined time envelope swaminathan 2011

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web the international occupational hygiene association ioaha refers to occupational hygiene as the discipline of anticipating recognizing evaluating and controlling health hazards in the working environment with the objective of protecting worker health and well being and safeguarding the community at large 4

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web occupational hygienists are committed to protecting the health and safety of people in the workplace and the community some occupational hygienists work in manufacturing petrochemical pharmaceutical steel mining and other industries others work in national governments hospitals and public utilities

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web occupational hygiene is a multi disciplinary science which means it draws on aspects of toxicology chemistry physics physiology maths and engineering to be able to assess workplace toxins such as dust noise vibration light and toxic gases

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web what is occupational hygiene occupational hygiene is the discipline of protecting worker health by controlling workplace hazards that can cause harm it also helps in maintaining worker well being and safeguarding the community at large

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web occupational hygiene is the science of preventing ill health from work activities it s practitioners come from varied backgrounds they can be chemists engineers biologists physicists doctors nurses and other professionals all of whom have chosen to apply their skills to protecting the health of workers