

# BALANCING REDOX EQUATIONS IN ACIDIC AND BASIC SOLUTION

Name _____	Section _____	Date _____
<b>Balancing Redox Equations in Acids and Basic Solution</b>		
<b>Section 1:</b> Balance each equation in acidic solution. Show the half reactions for oxidation and reduction.		
1. $\text{ZnO}_2 + \text{HClO}_2(\text{aq}) \rightarrow \text{Zn}^{2+}(\text{aq}) + \text{H}_2\text{O}(\text{l})$		
2. $\text{HNO}_2(\text{aq}) + \text{H}_2\text{O}_2(\text{aq}) \rightarrow \text{HNO}_3(\text{aq}) + \text{O}_2(\text{g}) + \text{H}^+(\text{aq})$		
3. $\text{ZnO}_2 + \text{Cr}_2\text{O}_7^{2-}(\text{aq}) \rightarrow \text{Cr}^{3+}(\text{aq}) + \text{Zn}^{2+}(\text{aq})$		
4. $\text{CrO}_2 + \text{SO}_3^{2-}(\text{aq}) \rightarrow \text{Cr}^{3+}(\text{aq}) + \text{SO}_4^{2-}(\text{aq})$		
5. $\text{NO}_2(\text{g}) + \text{HNO}_3(\text{aq}) \rightarrow \text{NO}_3^-(\text{aq}) + \text{H}^+(\text{aq})$		
6. $\text{FeO}_2(\text{aq}) + \text{SO}_3^{2-}(\text{aq}) \rightarrow \text{SO}_4^{2-}(\text{aq}) + \text{Fe}^{3+}(\text{aq})$		
7. $\text{I}_2(\text{s}) + \text{HClO}_2(\text{aq}) \rightarrow \text{IO}_3^-(\text{aq}) + \text{HClO}(\text{aq})$		
8. $\text{FeH}_2\text{O}_2(\text{aq}) + \text{H}_2\text{O}_2(\text{aq}) \rightarrow \text{Fe}^{3+}(\text{aq}) + \text{O}_2(\text{g})$		
9. $\text{Cr}_2\text{O}_7^{2-}(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{Cr}^{3+}(\text{aq}) + \text{Cl}_2(\text{g})$		
10. $\text{NO}_2(\text{aq}) + \text{NO}_2(\text{g}) \rightarrow \text{I}_2(\text{s}) + \text{NO}_3^-(\text{aq})$		

Name _____	Section _____	Date _____
<b>Section 2:</b> Balance each equation in basic solution. Show the half reactions for oxidation and reduction.		
11. $\text{Cl}^-(\text{aq}) + \text{Cr}_2\text{O}_7^{2-}(\text{aq}) \rightarrow \text{Cl}_2(\text{g}) + \text{Cr}^{3+}(\text{aq})$		
12. $\text{Mn}^{2+}(\text{aq}) + \text{O}_2(\text{g}) \rightarrow \text{MnO}_2(\text{s}) + \text{OH}^-(\text{aq})$		
13. $\text{Ce}^{4+}(\text{aq}) + \text{H}_2\text{O}_2 \rightarrow \text{Ce}(\text{OH})_3(\text{s}) + \text{H}_2\text{O}(\text{l})$		
14. $\text{Mn}^{2+}(\text{aq}) + \text{BrO}_3^-(\text{aq}) \rightarrow \text{MnO}_2(\text{s}) + \text{Br}^-(\text{aq})$		
15. $\text{HNO}_2(\text{aq}) + \text{MnO}_4^-(\text{aq}) \rightarrow \text{Mn}^{2+}(\text{aq}) + \text{NO}_3^-(\text{aq})$		
16. $\text{Cr}_2\text{O}_7^{2-}(\text{aq}) + \text{Cr}^{3+}(\text{aq}) \rightarrow \text{Cr}^{3+}(\text{aq}) + \text{Cr}^{3+}(\text{aq})$		
17. $\text{H}_2\text{Ag}_2 + \text{NO}_3^-(\text{aq}) \rightarrow \text{NO}_2(\text{g}) + \text{I}_2(\text{s})$		
18. $\text{MnO}_2(\text{aq}) + \text{I}_2\text{O}_5(\text{aq}) \rightarrow \text{Mn}^{2+}(\text{aq}) + \text{SO}_4^{2-}(\text{aq})$		
19. $\text{Cl}_2(\text{g}) \rightarrow \text{Cl}^-(\text{aq}) + \text{OCl}_2(\text{aq})$		
20. $\text{MnO}_2(\text{aq}) + \text{I}^-(\text{aq}) \rightarrow \text{MnO}_2(\text{s}) + \text{IO}_3^-(\text{aq})$		

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20 PROBLEMS

WORKED ANSWER KEY INCLUDED

# Solutions Manual Teacher Redox

**ML Morrison**



## **Solutions Manual Teacher Redox:**

## Unveiling the Energy of Verbal Artistry: An Psychological Sojourn through **Solutions Manual Teacher Redox**

In a world inundated with screens and the cacophony of fast interaction, the profound energy and psychological resonance of verbal art often fade into obscurity, eclipsed by the continuous barrage of sound and distractions. Yet, located within the lyrical pages of **Solutions Manual Teacher Redox**, a interesting perform of literary elegance that impulses with raw feelings, lies an unforgettable journey waiting to be embarked upon. Composed with a virtuoso wordsmith, that enchanting opus guides visitors on a psychological odyssey, lightly revealing the latent possible and profound impact stuck within the delicate internet of language. Within the heart-wrenching expanse with this evocative analysis, we will embark upon an introspective exploration of the book is main subjects, dissect their fascinating writing model, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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web 2<sup>a</sup> edición de esta obra dedicada a la cardiología aplicada al deporte cuyos objetivos son dos en primer lugar la identificación de las cardiopatías más frecuentes y su

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web kardiyoloji bölümünün baktığı birçok hastalık vardır göğüs bölgesindeki ağrı ödem çarpıntı ve nefes almada güçlük gibi sorunlar kardiyoloji alanı ile ilgilidir kalp hastalıkları ihmal

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web esta nueva edición de cardiología en el deporte completamente revisada y actualizada aborda los efectos del ejercicio físico sobre la salud cardiovascular tanto en el