

UNIVERSITY OF PUNE
[4361]-103
F. E. Examination - 2013
ENGINEERING CHEMISTRY
(2012 Pattern)

[Time : 2 Hours]

[Max. Marks : 50]

Instructions :

- (1) Answer Q1 or Q2, Q3 or Q4, Q5 or Q6, Q7 or Q8.
 - (2) Black figures to the right indicate full marks.
 - (3) Neat diagrams must be drawn wherever necessary.
 - (4) Use of logarithmic tables, slide rule, Mollier charts, electronics pocket calculator is allowed.
 - (5) Assume suitable data, if necessary.
-

Q1 a) Explain boiler corrosion and caustic embrittlement as ill effects of using hard water in boilers. State their causes and preventive measures. [6]

b) Which are possible transitions, that occur when molecule absorbs uv- visible radiation? Which type of electronic transitions will be possible in following molecules. [3]



c) State the reference electrode and standard electrode used in pH metery, potentiometry and conductometry. [3]

OR

Q2 a) 50 ml std. hard water containing 1.2 gm CaCO_3 per lit. required 15 ml EDTA solution for the end point. Whereas 50 ml sample water required 19ml of EDTA soln. and 50 ml boiled sample water required 11ml of EDTA soln. for the end point. Calculate, total, temporary and permanent hardness of sample water in ppm. [3]

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Table of Contents Solved Question Paper Pune University For Fe

1. Understanding the eBook Solved Question Paper Pune University For Fe
 - The Rise of Digital Reading Solved Question Paper Pune University For Fe
 - Advantages of eBooks Over Traditional Books
2. Identifying Solved Question Paper Pune University For Fe
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Solved Question Paper Pune University For Fe
 - User-Friendly Interface
4. Exploring eBook Recommendations from Solved Question Paper Pune University For Fe
 - Personalized Recommendations
 - Solved Question Paper Pune University For Fe User Reviews and Ratings
 - Solved Question Paper Pune University For Fe and Bestseller Lists
5. Accessing Solved Question Paper Pune University For Fe Free and Paid eBooks
 - Solved Question Paper Pune University For Fe Public Domain eBooks
 - Solved Question Paper Pune University For Fe eBook Subscription Services
 - Solved Question Paper Pune University For Fe Budget-Friendly Options
6. Navigating Solved Question Paper Pune University For Fe eBook Formats

- ePub, PDF, MOBI, and More
 - Solved Question Paper Pune University For Fe Compatibility with Devices
 - Solved Question Paper Pune University For Fe Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Solved Question Paper Pune University For Fe
 - Highlighting and Note-Taking Solved Question Paper Pune University For Fe
 - Interactive Elements Solved Question Paper Pune University For Fe
 8. Staying Engaged with Solved Question Paper Pune University For Fe
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Solved Question Paper Pune University For Fe
 9. Balancing eBooks and Physical Books Solved Question Paper Pune University For Fe
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Solved Question Paper Pune University For Fe
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Solved Question Paper Pune University For Fe
 - Setting Reading Goals Solved Question Paper Pune University For Fe
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Solved Question Paper Pune University For Fe
 - Fact-Checking eBook Content of Solved Question Paper Pune University For Fe
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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