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Chemistry 30

Thermochemistry Review

- 1. Which of the following statements is true for an endothermic reaction?
 - A. ΔH_e for the reaction is negative.
 - B. Energy is a product in the chemical equation.
 - C. The temperature of the surroundings increases.
 - D. The potential energy of the products is greater than the potential energy of the reactants.

Use the following information to answer the next question.

$$4 Q(s) + Z_2(g) + 100 kJ \rightarrow 2 Q_2Z(g)$$

- The ΔH_r value for the reverse reaction would be
 - A. +100 kJ
 - B. +50.0 kJ
 - C. -50.0 kJ
 - D. -100 kJ

Numerical Response

Benzene is a gasoline additive. The heat of formation for 1.00 mol of C₆H₆(1) is kJ.

(Record your three-digit answer in the numerical-response section on the answer sheet)

Numerical Response

An outdoor mercury thermometer contains 1.02 g of mercury. On a winter day, the thermometer records a change in temperature from −38.0°C to −13.0°C. The mercury absorbed ______ J of energy. The heat capacity of mercury is 0.140 J/g°C.

(Record your three-digit answer in the numerical-response section on the answer sheet)

Pearson Chemistry Section Review Answers Thermochemistry

Albert S. Tarendash

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has always believed the behavior of matter is determined by the properties of molecules and atoms to be the most important discovery in scientific knowledge This idea is the entire organizing factor for his seminal new text Chemistry Structure and Properties Dr Tro emphasizes the relationship between structure and properties establishes a unique atoms first approach to teaching chemistry by presenting atomic and bonding theories early in the text and stresses key themes throughout The book is organized to present chemistry as a logical cohesive story from the microscopic to the macroscopic so students can fully grasp the theories and framework behind the chemical facts Every topic has been carefully crafted to illustrate that the relationship between structure and properties is the thread that weaves all of chemistry together While developed independently of other Tro texts Chemistry Structure and Properties incorporates the author's vivid writing style chemical rigor dynamic multi level images and tested features His consistent conceptual focus and step by step problem solving framework encourages you to think through processes rather than simply memorize content Interactive media within MasteringChemistry complements the book s problem solving approach thus creating a comprehensive program that enables you to learn both in and out of the classroom This program presents a better teaching and learning experience for you Personalized learning with MasteringChemistry This online homework tutorial and assessment program is designed to improve results by helping you quickly master concepts You ll benefit from self paced tutorials featuring specific wrong answer feedback and hints that emulate the office hour experience Developed with a central theme and by a teaching community As part of a community that teaches with the understanding that matter is composed of particles and the structure of those particles determines the properties of matter Dr Tro took great lengths in the text to ensure that everything from organization art and pedagogy reinforce this theme The result of this emphasis is that the topic order has been constructed to make key connections earlier stronger and more often than the traditional approach Linking conceptual understanding with problem solving skills Throughout each chapter numerous Conceptual Connections encourage comprehension of the most complex concepts while a consistent step by step framework in the worked examples allows you to think logically through the problem solving process Visualizing and understanding chemistry Revolutionary multipart images illustrate and reinforce the theme of the text and allows you to see and experience the molecules responsible for the **Practice Makes Perfect Chemistry Review and Workbook, Second Edition** structures and properties of matter Marian DeWane, Heather Hattori, 2018-12-28 The Winning Equation for Success in Chemistry is Practice Practice This book will help you apply concepts and see how chemistry topics are interconnected Inside are numerous lessons to help you better understand the subject These lessons are accompanied by dozens of exercises to practice what you ve learned along with a complete answer key to check your work Throughout this book you will learn the terms to help you understand chemistry and you will expand your knowledge of the subject through hundreds of sample questions and their solutions With

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