

Thermal Energy, Heat, and Particle Motion (Worksheet)

Q.1. Which of the following statements is **false**?

- a) Matters are composed of tiny particles called atoms and molecules.
- c) The particles of matter are moving constantly.
- b) The thermal energy that is transferred from warmer things to cooler things is called heat.
- d) Heat and temperature are the same thing.

Q.2. Which one of the following terms is defined as the sum of the kinetic and potential energy of all the particles in a substance?

- a) Electricity
- b) Thermal energy
- c) Atom
- d) Molecule

Q. Use the image below to answer the next three questions (Q.3 - Q.4 - Q.5)



Q.3. Write **T** or **True** if the statement is true; write **F** or **False** if the statement is false.

- _____ 1. The particles in the water vapor move slower than those in the liquid water.
- _____ 2. There are spaces between the particles of matter.
- _____ 3. A change in thermal energy can lead to a change in phase.
- _____ 4. Adding or removing thermal energy doesn't lead to changes in particle motion.

Q.4. When a substance is heated, the space between the particles of the substance _____

- a) decreases.
- b) does not change.
- c) increases.

Q.5. When a substance is heated, the particles of the substance move _____

- a) slower.
- b) faster.

Pearson Education Thermal Energy And Heat Answers

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