

2

MATTER AND CHANGE

SECTION 2.1 PROPERTIES OF MATTER (pages 39–42)

This section helps you distinguish extensive from intensive properties and identify substances by their properties. It teaches you how to differentiate the three states of matter. It also defines a physical property and lists examples of physical properties and physical changes.

► Describing Matter (page 39)

1. The _____ mass _____ of an object is a measure of the amount of matter the object contains.

2. How does an extensive property differ from an intensive property?

An extensive property depends on the amount of matter, while an intensive property depends on the type of matter.

► Identifying Substances (page 40)

3. Matter that has a uniform and definite composition is called a _____ substance _____.

4. Is the following sentence true or false? All samples of a substance have different physical properties. _____ false _____

5. A physical property is a quality or condition of a substance that can be _____ observed _____ or _____ measured _____ without changing the substance's composition.

6. Circle the letter of the term that is NOT a physical property.

- a. hardness c. boiling point
b. color d. melting

7. Look at Table 2.1 on page 40. What is the melting point of bromine? _____ -7°C _____

8. Look at Table 2.1 on page 40. Circle the letter of the substance that is a yellow solid and melts at 115°C.

- a. sulfur
b. chlorine
c. gold
d. copper

Pearson Education Changes In Matter Answer Key

Victor M. Corman



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1. Understanding the eBook Pearson Education Changes In Matter Answer Key
 - The Rise of Digital Reading Pearson Education Changes In Matter Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Education Changes In Matter Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pearson Education Changes In Matter Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Education Changes In Matter Answer Key

- Personalized Recommendations
- Pearson Education Changes In Matter Answer Key User Reviews and Ratings
- Pearson Education Changes In Matter Answer Key and Bestseller Lists
- 5. Accessing Pearson Education Changes In Matter Answer Key Free and Paid eBooks
 - Pearson Education Changes In Matter Answer Key Public Domain eBooks
 - Pearson Education Changes In Matter Answer Key eBook Subscription Services
 - Pearson Education Changes In Matter Answer Key Budget-Friendly Options
- 6. Navigating Pearson Education Changes In Matter Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Pearson Education Changes In Matter Answer Key Compatibility with Devices
 - Pearson Education Changes In Matter Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pearson Education Changes In Matter Answer Key
 - Highlighting and Note-Taking Pearson Education Changes In Matter Answer Key
 - Interactive Elements Pearson Education Changes In Matter Answer Key
- 8. Staying Engaged with Pearson Education Changes In Matter Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pearson Education Changes In Matter Answer Key
- 9. Balancing eBooks and Physical Books Pearson Education Changes In Matter Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pearson Education Changes In Matter Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Pearson Education Changes In Matter Answer Key
 - Setting Reading Goals Pearson Education Changes In Matter Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pearson Education Changes In Matter Answer Key

- Fact-Checking eBook Content of Pearson Education Changes In Matter Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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