

# Temperature Mix

## Purpose

To predict the final temperature of a mixture of cups of water at different temperatures

## Apparatus

3 Styrofoam cups  
liter container with a wide mouth  
thermometer (Celsius)  
pail of cold water  
pail of hot water

## Discussion

If you mix a pail of cold water with a pail of hot water, the final temperature of the mixture will be between the two initial temperatures. What information would you need to predict the final temperature? You'll begin with the simplest case of mixing **equal** masses of hot and cold water.



## Procedure

**Step 1:** Begin by marking your three Styrofoam cups equally at about the three-quarter mark. You can do this by pouring water from one container to the next and mark the levels along the inside of each cup.

**Step 2:** Fill the first cup to the mark with hot water from the pail, and fill the second cup with cold water to the same level. Measure and record the temperature of both cups of water.

Temperature of cold water = 23 °C

Temperature of warm water = 39 °C

**Step 3:** Predict the temperature of the water when the two cups are combined. Then pour the two cups of water into the liter container, stir the mixture slightly, and record its temperature.

Predicted temperature = 31 °C

Actual temperature of water = 31 °C



If there was a difference between your prediction and your observation, what may have caused it?

Pour the mixture into the sink or waste pail. (Don't be a klutz and pour it back into either of the pails of cold or hot water!) Now you'll investigate what happens when **unequal** amounts of hot and cold water are combined.

# Temperature Mix Lab Manual Physics

**CO Houle**



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**Nuclear Science Abstracts**, 1976      **A Laboratory Manual of Foods and Cookery** Emma B. Matteson, Ethel M. Newlands, 1916      **Atomic Energy Research Reports for Sale by the U.S. Department of Commerce, Office of Technical Services** Business and Defense Services Administration, 1963      **Scientific and Technical Aerospace Reports**, 1995      **The Chemical News**, 1871      The Nation, 1894      List, 1948      *Nature* Sir Norman Lockyer, 1896      **The Middle Paraná River** Martin H. Iriondo, Juan César Paggi, María Julieta Parma, 2007-06-03 1 2 2 3 MART N H IRIONDO JUAN C SAR PAGGI AND MAR A JULIETA PARMA 2 The great Paran basin 2 600 000 km covers a large portion of the Neotropical Realm extending from the Andes to the Atlantic coast and c necting the central regions of South America along 2 200 km in a north south direction Most of the basin is under subtropical climates which range from desertic in the west to humid in the east The Middle Paran is a broad complex floodplain extending 600 km in the heart of the interior lowlands of the continent This system of river channels shallow lakes islands and wetlands remains unoccupied by man and basically unaltered From a systemic point of view this area can be considered to be the most important link and collector of the fluvial chain that conveys huge volumes of nutrients sediments and salts from the continent to the South Atlantic Ocean The influence of the Paran upon the Atlantic is very important It has been 2 calculated that a sediment deposit 400 m thick and 1 million km in area ac mulated in the platform and nearby oceanic bottom over the last 2 million years with a similar amount of dissolved salts contributed by the river to the ocean This book details the recent advances in the knowledge of the most important segment of the Parana River organized according to the scientific structure of the National Institute of Limnology of Argentina INALI CONICET UNL      **INIS Atomindex**, 1984      **A Laboratory Manual of General Chemistry** Walter Scott Hendrixon, 1918      Selected Water Resources Abstracts, 1981

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




























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