

Periodic Table

Main groups

1A ^a												Main groups						
1																		
1	<div><div>1</div><div>H</div><div>1.00794</div></div>	2A											3A	4A	5A	6A	7A	8A
	2												13	14	15	16	17	18
2	<div><div>3</div><div>Li</div><div>6.941</div></div>	<div><div>4</div><div>Be</div><div>9.01218</div></div>	Transition metals										<div><div>5</div><div>B</div><div>10.811</div></div>	<div><div>6</div><div>C</div><div>12.0107</div></div>	<div><div>7</div><div>N</div><div>14.0067</div></div>	<div><div>8</div><div>O</div><div>15.9994</div></div>	<div><div>9</div><div>F</div><div>18.9984</div></div>	<div><div>10</div><div>Ne</div><div>20.1797</div></div>
3	<div><div>11</div><div>Na</div><div>22.98976</div></div>	<div><div>12</div><div>Mg</div><div>24.3050</div></div>	3B	4B	5B	6B	7B	8B			1B	2B	<div><div>13</div><div>Al</div><div>26.9815</div></div>	<div><div>14</div><div>Si</div><div>28.0855</div></div>	<div><div>15</div><div>P</div><div>30.9738</div></div>	<div><div>16</div><div>S</div><div>32.065</div></div>	<div><div>17</div><div>Cl</div><div>35.453</div></div>	<div><div>18</div><div>Ar</div><div>39.948</div></div>
	3	4	3	4	5	6	7	8	9	10	11	12						
4	<div><div>19</div><div>K</div><div>39.0983</div></div>	<div><div>20</div><div>Ca</div><div>40.078</div></div>	<div><div>21</div><div>Sc</div><div>44.9559</div></div>	<div><div>22</div><div>Ti</div><div>47.867</div></div>	<div><div>23</div><div>V</div><div>50.9415</div></div>	<div><div>24</div><div>Cr</div><div>51.9961</div></div>	<div><div>25</div><div>Mn</div><div>54.9380</div></div>	<div><div>26</div><div>Fe</div><div>55.845</div></div>	<div><div>27</div><div>Co</div><div>58.9332</div></div>	<div><div>28</div><div>Ni</div><div>58.6934</div></div>	<div><div>29</div><div>Cu</div><div>63.546</div></div>	<div><div>30</div><div>Zn</div><div>65.39</div></div>	<div><div>31</div><div>Ga</div><div>69.723</div></div>	<div><div>32</div><div>Ge</div><div>72.64</div></div>	<div><div>33</div><div>As</div><div>74.9214</div></div>	<div><div>34</div><div>Se</div><div>78.96</div></div>	<div><div>35</div><div>Br</div><div>79.904</div></div>	<div><div>36</div><div>Kr</div><div>83.80</div></div>
5	<div><div>37</div><div>Rb</div><div>85.4678</div></div>	<div><div>38</div><div>Sr</div><div>87.62</div></div>	<div><div>39</div><div>Y</div><div>88.9059</div></div>	<div><div>40</div><div>Zr</div><div>91.224</div></div>	<div><div>41</div><div>Nb</div><div>92.9064</div></div>	<div><div>42</div><div>Mo</div><div>95.94</div></div>	<div><div>43</div><div>Tc</div><div>[98]</div></div>	<div><div>44</div><div>Ru</div><div>101.07</div></div>	<div><div>45</div><div>Rh</div><div>102.9055</div></div>	<div><div>46</div><div>Pd</div><div>106.42</div></div>	<div><div>47</div><div>Ag</div><div>107.8682</div></div>	<div><div>48</div><div>Cd</div><div>112.411</div></div>	<div><div>49</div><div>In</div><div>114.818</div></div>	<div><div>50</div><div>Sn</div><div>118.710</div></div>	<div><div>51</div><div>Sb</div><div>121.760</div></div>	<div><div>52</div><div>Te</div><div>127.60</div></div>	<div><div>53</div><div>I</div><div>126.9045</div></div>	<div><div>54</div><div>Xe</div><div>131.293</div></div>
6	<div><div>55</div><div>Cs</div><div>132.9055</div></div>	<div><div>56</div><div>Ba</div><div>137.327</div></div>	<div><div>71</div><div>Lu</div><div>174.967</div></div>	<div><div>72</div><div>Hf</div><div>178.49</div></div>	<div><div>73</div><div>Ta</div><div>180.9479</div></div>	<div><div>74</div><div>W</div><div>183.84</div></div>	<div><div>75</div><div>Re</div><div>186.207</div></div>	<div><div>76</div><div>Os</div><div>190.23</div></div>	<div><div>77</div><div>Ir</div><div>192.217</div></div>	<div><div>78</div><div>Pt</div><div>195.078</div></div>	<div><div>79</div><div>Au</div><div>196.9666</div></div>	<div><div>80</div><div>Hg</div><div>200.59</div></div>	<div><div>81</div><div>Tl</div><div>204.3833</div></div>	<div><div>82</div><div>Pb</div><div>207.2</div></div>	<div><div>83</div><div>Bi</div><div>208.9804</div></div>	<div><div>84</div><div>Po</div><div>[209]</div></div>	<div><div>85</div><div>At</div><div>[209]</div></div>	<div><div>86</div><div>Rn</div><div>[222]</div></div>
7	<div><div>87</div><div>Fr</div><div>[223]</div></div>	<div><div>88</div><div>Ra</div><div>[226]</div></div>	<div><div>103</div><div>Lr</div><div>[262]</div></div>	<div><div>104</div><div>Rf</div><div>[261]</div></div>	<div><div>105</div><div>Db</div><div>[262]</div></div>	<div><div>106</div><div>Sg</div><div>[266]</div></div>	<div><div>107</div><div>Bh</div><div>[264]</div></div>	<div><div>108</div><div>Hs</div><div>[269]</div></div>	<div><div>109</div><div>Mt</div><div>[268]</div></div>	<div><div>110</div><div>Ds</div><div>[271]</div></div>	<div><div>111</div><div>Rg</div><div>[272]</div></div>	<div><div>112</div><div>Cn</div><div>[277]</div></div>	<div><div>113</div><div></div><div>[284]</div></div>	<div><div>114</div><div></div><div>[289]</div></div>	<div><div>115</div><div></div><div>[288]</div></div>	<div><div>116</div><div></div><div>[292]</div></div>		

Lanthanide series	57 La 138.9055	58 Ce 140.116	59 Pr 140.9077	60 Nd 144.24	61 Pm [145]	62 Sm 150.36	63 Eu 151.964	64 Gd 157.25	65 Tb 158.9253	66 Dy 162.50	67 Ho 164.9303	68 Er 167.259	69 Tm 168.9342	70 Yb 173.04
Actinide series	89 Ac [227]	90 Th 232.0381	91 Pa 231.0369	92 U 238.0289	93 Np [237]	94 Pu [244]	95 Am [243]	96 Cm [247]	97 Bk [247]	98 Cf [251]	99 Es [252]	100 Fm [257]	101 Md [258]	102 No [259]

^aThe labels on top (1A, 2A, etc.) are common American usage. The labels below these (1, 2, etc.) are those recommended by the International Union of Pure and Applied Chemistry (IUPAC).

Pearson Chemistry Workbook Periodic Table

Yicheng Fang



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