

Chapter 8 Answers

Practice 8-1

- 1:278
2. 18 ft by 10 ft
3. 18 ft by 14 ft
4. 18 ft by 16 ft
5. 8 ft by 8 ft
6. 3 ft by 10 ft
7. true
8. false
9. true
10. true
11. false
12. true
13. true
14. false
15. false
16. 12
17. 12
18. 33
19. $\frac{21}{2}$
20. ± 8
21. $\frac{49}{2}$
22. 6
23. $\frac{4}{3}$
24. $-\frac{4}{3}$
25. 14:5
26. 3:2
27. 12:7
28. $\frac{4}{3}$
29. $\frac{7}{13}$

Practice 8-2

1. $\triangle ABC \sim \triangle XYZ$, with similarity ratio 2:1
2. $\triangle QMN \sim \triangle RST$, with similarity ratio 5:3
3. Not similar; corresponding sides are not proportional.
4. Not similar; corresponding angles are not congruent.
5. $\triangle ABC \sim \triangle KMN$, with similarity ratio 4:7
6. Not similar; corresponding sides are not proportional.
7. $\angle I$
8. $\angle O$
9. $\angle J$
10. NO
11. LO
12. LO
13. 3.96 ft
14. 4.8 in.
15. 3.75 cm
16. 10 m
17. $\frac{4}{3}$
18. 53
19. $7\frac{1}{2}$
20. $4\frac{1}{2}$
21. 55
22. 37
23. 5

Practice 8-3

1. $\angle AXB = \angle RXQ$ because vertical angles are \cong . $\angle A = \angle R$ (Given). Therefore $\triangle AXB \sim \triangle RXQ$ by the AA \sim Postulate.
2. Because $\frac{MP}{LN} = \frac{PN}{LN} = \frac{2}{5}$, $\frac{PM}{LN} = \frac{2}{5}$. $\triangle MPX \sim \triangle LWA$ by the SSS \sim Theorem.
3. $\angle QMP \cong \angle AMB$ because vertical \angle s are \cong . Then, because $\frac{QM}{LM} = \frac{PM}{LM} = \frac{1}{2}$, $\triangle QMP \sim \triangle AMB$ by the SAS \sim Theorem.
4. $\angle M = \angle A$ (Given). Because there are 180° in a triangle, $m\angle I = 130$, and $\angle I = \angle C$. So $\triangle MIN \sim \triangle ACB$ by the AA \sim Postulate.
5. Because $AX = BX$ and $CX = CX$, $\frac{AX}{CX} = \frac{BX}{CX}$, $\angle AXB = \angle CXR$ because vertical angles are \cong . Therefore $\triangle AXB \sim \triangle CXR$ by the SAS \sim Theorem.
6. Because $AB = BC = CA$ and $XY = YZ = ZX$, $\frac{AB}{XY} = \frac{BC}{YZ} = \frac{CA}{ZX}$. Then $\triangle ABC \sim \triangle XYZ$ by the SSS \sim Theorem.
7. $\frac{1}{2}$
8. $\frac{22}{3}$
9. $\frac{45}{2}$
10. $\frac{49}{2}$
11. $\frac{49}{2}$
12. 36
13. 33 ft

Practice 8-4

1. 16
2. 8
3. $\sqrt{77}$
4. $2\sqrt{11}$
5. $10\sqrt{2}$
6. $6\sqrt{3}$
7. b
8. y
9. x
10. a
11. b
12. c
13. $x = 6$; $y = 6\sqrt{3}$
14. $x = 8\sqrt{3}$; $y = 4\sqrt{3}$
15. $\frac{40\sqrt{105}}{29}$
16. $\frac{2}{3}$
17. $x = 4\sqrt{5}$; $y = \sqrt{55}$
18. $x = \frac{45}{2}$; $y = \frac{27}{2}$
19. $x = \sqrt{3}$; $y = \sqrt{6}$; $z = \sqrt{2}$
20. $x = \frac{3}{2}$; $y = \frac{1}{2}$; $z = \frac{1}{2}$
21. $x = 8$; $y = 2\sqrt{2}$; $z = 6\sqrt{2}$
22. $2\sqrt{15}$ in.

Practice 8-5

1. BE
2. EH
3. BC
4. JD
5. $\frac{49}{2}$
6. BE
7. $\frac{49}{2}$
8. 4
9. 4
10. $\frac{49}{2}$
11. $x = \frac{25}{2}$; $y = 4$
12. $\frac{12}{5}$
13. $x = \frac{45}{2}$; $y = \frac{15}{2}$
14. $x = 6$; $y = 6$
15. $x = \frac{140}{3}$; $y = \frac{140}{3}$
16. 2
17. 4
18. 10

Practice 8-6

1. 4:5; 16:25
2. 5:3; 25:9
3. 3:4; 9:16
4. 1:2
5. 9:5
6. 1:6
7. 8:3
8. $\frac{225}{2}$ in.²
9. $\frac{2250}{2}$ cm²
10. $\frac{1152}{5}$ in.²
11. 2:3
12. 1:2
13. 2:5
14. 108 ft²

Reteaching 8-1

1. 534 apples
2. 768 gallons
3. 4627 employees
4. about 4800 picture frames
5. 90
6. 2450
7. 85
8. 560
9. 58
10. 7308
11. 12
12. 20,250
13. 2500

Reteaching 8-2

1. $\triangle KLM \sim \triangle QRS$; 5:3
2. $\triangle ABC \sim \triangle XYZ$; 10:7
3. not similar
4. not similar
5. $\triangle MPR \sim \triangle QAC$; 1:1
6. not similar
7. similar with ratio 10:11
8. not similar
9. similar with ratio 4:9

Reteaching 8-3

1. $\triangle ABC \sim \triangle ZYX$ by AA \sim
2. not similar
3. $\triangle QEU \sim \triangle SJO$ by SSS \sim
4. $\triangle ABC \sim \triangle RST$ by AA \sim
5. not similar
6. $\triangle BAC \sim \triangle XQR$ by SAS \sim
7. yes, by SSS \sim or by AA \sim
8. No; the vertex angles may differ in measure.
9. All congruent triangles are similar with ratio 1:1. Similar triangles are not necessarily congruent.

Reteaching 8-4

1. $x = \frac{25}{2}$; $y = \frac{145}{2}$; $z = \frac{40}{2}$
2. $x = \frac{400}{29}$; $y = \frac{440}{29}$
3. $x = 48$; $y = 8$
4. $x = 12$; $y = 16$
5. $x = \sqrt{2}$; $y = \sqrt{3}$; $z = \sqrt{6}$
6. $x = 2\sqrt{10}$; $y = \sqrt{15}$; $z = 2\sqrt{6}$
7. $\frac{120}{13}$
8. $\frac{45}{\sqrt{85}} = \frac{45\sqrt{85}}{85}$
9. $\sqrt{\frac{40}{11}}$

Reteaching 8-5

1. $x = 45$
2. $x = \frac{5}{2}$
3. $x = 7.2$; $y = 4.8$
4. $x = \frac{12}{5}$; $y = \frac{20}{5}$
5. $x = 9$
6. $x = \frac{3}{2}$; $y = \frac{3}{2}$
7. $BN = \frac{10}{3}$; $CN = \frac{20}{3}$
8. $BY = \frac{4}{3}$

Reteaching 8-6

1. 4:3; 16:9
2. 8:5; 64:25
3. 9:3; 81:25
4. 4:3
5. 4:5
6. 3:5
7. 16:9
8. 16:25
9. 9:25

Pearson Education Chapter 1stoichiometry Answers

**H. Scott Fogler, Bryan R.
Goldsmith, Eranda Nikolla, Nirala Singh**

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Chemistry Pearson Education, Lemay, 2005-05 This lecture notebook contains the art from the text with note taking sections to obviate the need for students to spend time redrawing figures in lecture and instead lets them concentrate on taking notes

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Chemistry: The Central Science Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller *Chemistry The Central Science* An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation

Students Study Guide Fredeen, Pearson, 2003-07 IIT Chemistry: Introductory Topics , **The Software Encyclopedia** , 1988 **Elements of Chemical Reaction Engineering** H. Scott Fogler, Bryan R. Goldsmith, Eranda Nikolla, Nirala Singh, 2025-03-19 The Essential Textbook for Mastering Chemical Reaction Engineering Now Fully Updated with Expanded Coverage of Electrochemical Reactors H Scott Fogler's *Elements of Chemical Reaction Engineering* now in its seventh edition continues to set the standard as the leading textbook in chemical reaction engineering This edition coauthored by Bryan R Goldsmith Eranda Nikolla and Nirala Singh still offers Fogler's engaging and active

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Inorganic Chemistry Catherine E. Housecroft, A. G. Sharpe, 2005 Inorganic Chemistry Catherine E Housecroft and Alan G Sharpe This book has established itself as a leading textbook in the subject by offering a fresh and exciting approach to the teaching of modern inorganic chemistry It gives a clear introduction to key principles with strong coverage of descriptive chemistry of the elements Special selected topics chapters are included covering inorganic kinetics and mechanism catalysis solid state chemistry and bioinorganic chemistry A new full colour text design and three dimensional illustrations bring inorganic chemistry to life Topic boxes have been used extensively throughout the book to relate the chemistry described in the text to everyday life the chemical industry environmental issues and legislation and natural resources Teaching aids throughout the text have been carefully designed to

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Essentials of Chemical Reaction Engineering H. Scott Fogler, 2017-10-26 Today's Definitive Undergraduate Level Introduction to Chemical Reaction Engineering Problem Solving For 30 years H Scott Fogler's Elements of Chemical Reaction Engineering has been the 1 selling text for courses in chemical reaction engineering worldwide Now in Essentials of Chemical Reaction Engineering Second Edition Fogler has distilled this classic into a modern introductory level guide specifically for undergraduates This is the ideal resource for today's students learners who demand instantaneous access to information and want to enjoy learning as they deepen their critical thinking and creative problem solving skills Fogler successfully integrates text visuals and computer simulations and links theory to practice through many relevant examples This updated second edition covers mole balances conversion and reactor sizing rate laws and stoichiometry isothermal reactor design rate data collection analysis multiple reactions reaction mechanisms pathways bioreactions and bioreactors catalysis catalytic reactors nonisothermal reactor designs and more Its multiple improvements include a new discussion of activation energy molecular simulation and stochastic modeling and a significantly revamped chapter on heat effects in chemical reactors To promote the transfer of key skills to real life settings Fogler presents three styles of problems Straightforward problems that reinforce the principles of chemical reaction engineering Living Example

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of many undesired effects These papers provide insight into the variables controlling clay behaviour **Math Review**
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Table of Contents Pearson Education Chapter 1stoichiometry Answers

1. Understanding the eBook Pearson Education Chapter 1stoichiometry Answers
 - The Rise of Digital Reading Pearson Education Chapter 1stoichiometry Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Education Chapter 1stoichiometry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pearson Education Chapter 1stoichiometry Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Education Chapter 1stoichiometry Answers
 - Personalized Recommendations
 - Pearson Education Chapter 1stoichiometry Answers User Reviews and Ratings
 - Pearson Education Chapter 1stoichiometry Answers and Bestseller Lists
5. Accessing Pearson Education Chapter 1stoichiometry Answers Free and Paid eBooks
 - Pearson Education Chapter 1stoichiometry Answers Public Domain eBooks
 - Pearson Education Chapter 1stoichiometry Answers eBook Subscription Services
 - Pearson Education Chapter 1stoichiometry Answers Budget-Friendly Options

6. Navigating Pearson Education Chapter 1stoichiometry Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Pearson Education Chapter 1stoichiometry Answers Compatibility with Devices
 - Pearson Education Chapter 1stoichiometry Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pearson Education Chapter 1stoichiometry Answers
 - Highlighting and Note-Taking Pearson Education Chapter 1stoichiometry Answers
 - Interactive Elements Pearson Education Chapter 1stoichiometry Answers
8. Staying Engaged with Pearson Education Chapter 1stoichiometry Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pearson Education Chapter 1stoichiometry Answers
9. Balancing eBooks and Physical Books Pearson Education Chapter 1stoichiometry Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pearson Education Chapter 1stoichiometry Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pearson Education Chapter 1stoichiometry Answers
 - Setting Reading Goals Pearson Education Chapter 1stoichiometry Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pearson Education Chapter 1stoichiometry Answers
 - Fact-Checking eBook Content of Pearson Education Chapter 1stoichiometry Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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