

**Radioactive Dating****Warm-up Activity**

Circle correct statements.

**Problem 1** Read the passage. Then, answer the questions that follow.**Radioactive Dating**

Scientists often determine the age of things by measuring the amount of a substance that has decayed from a fixed quantity. This method is called **radioactive dating**. The substance that is fixed over a substantial period is **radioactive substance** or **radioactive isotope**, or **radioisotope**. These things are substances that lose the substance, known as **decay** or **radioactive decay**, over time and naturally half-life. A half-life is the amount of time needed for one-half of the radioactive substance to decay. The amount of substance of a particular radioactive element is a fixed, for example, 100 grams. In the process of time, there is decay in substance. The amount left after time

1. What is radioactive dating? \_\_\_\_\_
- \_\_\_\_\_
2. What is half-life? \_\_\_\_\_
- \_\_\_\_\_

**Problem 2** The table below lists the half-lives of certain radioactive elements. Use the table to answer the questions that follow.

Radioactive Elements	
Radioactive Element	Half-life
Carbon-14	5730 years
Uranium-238	4.5 billion years
Plutonium-239	24,100 years
Cesium-137	30 years

1. What is the half-life of uranium-238? \_\_\_\_\_
2. Suppose a fixed substance was half as much as when it was first created. How old is the fixed? \_\_\_\_\_
3. Suppose a fixed substance was fourth as much as when it was first created. How old is the fixed substance? \_\_\_\_\_
4. If 10 grams of substance 137 are present now, how much will be left in 30 years? How much is 1 billion years? \_\_\_\_\_
5. If 10 grams of substance 239 are present now, how much will be left in 24,100 years? \_\_\_\_\_

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**Rex M. Heyworth**



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