

Chemistry

Unit 2

Practice Exam 1 Answers

Section A—Multiple-choice questions

Question 1

- B** The high electronegativity difference between hydrogen and oxygen means that the O—H bond in water is highly polar, enabling hydrogen bonds to be formed between nearby water molecules.

Sulfur is less electronegative than oxygen. The polarity of the H—S bond in H_2S is much less than the polarity of the O—H bond in water. The attraction between H_2S molecules is described as a dipole-dipole interaction.

The bond between hydrogen and Te is non-polar. The only interactions between molecules of H_2Te are dispersion forces brought about through the formation of instantaneous dipoles.

Question 2

- D** The formation of a solution involves breaking bonds between solute particles, breaking intermolecular bonds between solvent molecules, and the formation of bonds between solvent molecules and solute particles.

Question 3

- C** $n(\text{Na}_2\text{CO}_3) = cV = 0.150 \times \frac{300}{1000} = 0.0450 \text{ mol}$
 $m(\text{Na}_2\text{CO}_3) = mM = 0.0450 \times 106.0 = 4.77 \text{ g}$

Question 4

- C** Water is a poor conductor of electricity due to the small number of hydronium ions and hydroxide ions that are produced by the self-ionisation reaction of water.

Question 5

- A** Nitric acid, HNO_3 , is a strong acid because it is completely ionised in water. Ethanoic acid, CH_3COOH , is a weak acid as it is partly ionised in water. A 0.1 M solution is considered to be dilute and 6 M concentrated.

Question 6

- B** When the acid H_2PO_4^- donates a proton, it forms the conjugate base HPO_4^{2-} .

Question 7

- A** pH is defined as $\text{pH} = -\log_{10}[\text{H}_3\text{O}^+]$. Since HCl is a strong acid and completely ionised in solution, it can be assumed that in each beaker $[\text{HCl}] = [\text{H}_3\text{O}^+]$. pH is only dependent on the $[\text{H}_3\text{O}^+]$, not the volume of solution. Because the definition involves the negative logarithm, the pH increases as the $[\text{H}_3\text{O}^+]$ decreases. The solution that has the lowest $[\text{H}_3\text{O}^+]$ will have the highest pH.

Pearson General Chemistry Answers

Karen Timberlake



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Master the PCAT Peterson's, 2012-07-15 Peterson's Master the PCAT is an in depth review that offers thorough preparation for the computer based exam After learning about the structure format scoring and score reporting and the subtests and question types you can take a diagnostic test to learn about your strengths and weaknesses The next six parts of the eBook are focused on detailed subject reviews for each subtest verbal ability reading comprehension biology chemistry quantitative ability and writing Each review includes practice questions with detailed answer explanations You can take two practice tests to track your study progress The tests also offer detailed answer explanations to further improve your knowledge and understanding of the tested subjects The eBook concludes with an appendix that provides helpful information on a variety of careers in pharmacy and ten in depth career profiles

Examination Questions and Answers in Basic Anatomy and Physiology Martin Caon, 2018-04-06 This second edition provides 2400 multiple choice questions on human anatomy and physiology and some physical science separated into 40 categories The answer to each question is accompanied by an explanation Each category has an introduction to set the scene for the questions to come However not all possible information is provided within these Introductions so an Anatomy and Physiology textbook is an indispensable aid to understanding the answers The questions have been used in end of semester examinations for undergraduate anatomy and

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Learning and Collaboration Technologies: Games and Virtual Environments for Learning Panayiotis

Zaphiris,Andri Ioannou,2021-07-03 This two volume set LNCS 12784 and 12785 constitutes the refereed proceedings of the 8th International Conference on Learning and Collaboration Technologies LCT 2021 held as Part of the 23rd International Conference HCI International 2021 which took place in July 2021 Due to COVID 19 pandemic the conference was held virtually The total of 1276 papers and 241 posters included in the 39 HCII 2021 proceedings volumes was carefully reviewed and selected from 5222 submissions The papers of LCT 2021 Part II focus on Games and Gamification in Learning Chatbots in Learning AR VR and Robots in Learning *Organic Chemistry Education Research into Practice* Jay Wackerly,Sarah Zingales,Michael Wentzel,Gautam Bhattacharyya,Brett McCollum,2025-03-25 This Research Topic has three main goals 1 provide a platform for instructors of organic chemistry to showcase evidence based methods and educational theories they have utilized in their classrooms 2 build new and strengthen existing connections between educational researchers and practitioners and 3 highlight how people have used chemical education based research in their teaching practice There are places in the literature dedicated for chemical education research CER however there is not a clear avenue for those that have changed their teaching methods based on published CER and report their experiences Creating this article collection will foster collaboration between chemical education researchers and teachers of organic chemistry This opportunity allows these instructors to share evidence based practices experiences challenges and innovative approaches from CER literature and beyond This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes

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