

Balancing & Reactions Test SCH3U

Name: _____

 K T C A Assessment of Learning (Product)
10 20 6 7

Multiple Choice: Circle the most correct answer. (10K)

1. Classify the following chemical reaction: $2\text{AsCl}_3 + 3\text{H}_2\text{S} \rightarrow \text{As}_2\text{S}_3 + 6\text{HCl}$

- A) combustion B) synthesis C) decomposition
D) single displacement E) double displacement

2. The type of reaction that takes place when one element reacts with a compound to form a new compound and a different element is a _____.

- A) synthesis reaction C) single-displacement reaction
B) decomposition reaction D) double-displacement reaction

3. Which of the following types of chemical reaction produces water and carbon dioxide gas?

- A) single displacement B) double displacement C) decomposition D) combustion

4. When the equation $\text{NH}_4\text{OH} + \text{H}_2\text{SO}_4 \rightarrow (\text{NH}_4)_2\text{SO}_4 + \text{H}_2\text{O}$ is balanced, one term is:

- A) NH_4SO_4 B) 2 H_2O C) H_2OH D) 2 $(\text{NH}_4)_2\text{SO}_4$

5. When the eq. $\text{H}_2 + \text{V}_2\text{O}_5 \rightarrow \text{V} + \text{H}_2\text{O}$ is completed and balanced, one term is:

- A) 2 V_2O_5 B) 3 H_2O C) 2 V D) 8 H_2

6. Boron trifluoride reacts with water to produce two products, H_3BO_3 and HBF_4 . Which of the following equations correctly represents this reaction?

- A) $3\text{BF}_3 + 3\text{H}_2\text{O} \rightarrow \text{H}_3\text{BO}_3 + \text{HBF}_4$
B) $2\text{BF}_3 + 2\text{H}_2\text{O} \rightarrow 2\text{H}_3\text{BO}_3 + \text{HBF}_4$
C) $4\text{BF}_3 + 3\text{H}_2\text{O} \rightarrow \text{H}_3\text{BO}_3 + 3\text{HBF}_4$
D) $4\text{BF}_3 + 3\text{H}_2\text{O} \rightarrow \text{H}_3\text{BO}_3 + 3\text{HBF}_4$

7. In a double-displacement reaction, the _____.

- A) products are always molecular B) reactants are usually two ionic compounds
C) reactants are two elements D) products are a new element and a new compound

8. Which of the following statements is true about single-displacement reactions?

- A) They are restricted to metals. C) Two reactants produce two products.
B) They involve a single product. D) Any metal replaces any other metal.

9. Some acids, such as carbonic acid, decompose to nonmetal oxides and

- A) water B) a salt C) oxygen D) peroxide

Pearson Chemical Reactions Test Answers Chemistry

John Gormally



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