

6. The new penny is lighter because it is mostly zinc which has a lower density than copper.

You're The Chemist

- at 90°, mass of 1 drop: 0.019 g
at 45°, mass of 1 drop: 0.0218 g
at 0°, mass of 1 drop: 0.0242 g
Pipets give different results.
- The best angle is 90° because the pipet is easiest to control. Expel the air bubble so that the first drop will be the same size as the rest.
- Find mass of can and divide by density of aluminum. Sample answer: mass of one can: 14.77 g;
 d_{Al} : 2.70 g/cm³; $V = 5.47 \text{ cm}^3$
- (1) Measure the mass before and after you fill the can with water. Use the mass and density of water to find the volume.
(2) Measure the height and radius and calculate volume.
 $V = \pi r^2 h$
(Can is not a perfect cylinder.)
(3) Read label: 12 oz = 355 mL.
- Sample answer:
 $V = 16.5 \text{ m} \times 3.0 \text{ m} \times 12.8 \text{ m}$
 $= 630 \text{ m}^3 \times 1000 \text{ L/m}^3$
 $= 630,000 \text{ L}$
Assume 30 people with an average weight of 130 lb (1 kg = 2.2 lb) and a density of about 1.0 kg/L.
Volume of 30 people:
 $V = 30 \times 130 \text{ lb} \times 1 \text{ kg}/2.2 \text{ lb} \times 1 \text{ L}/1.0 \text{ kg}$
 $= 1800 \text{ L}$
The volume of 30 chairs, 15 tables, and 2 desks is about that of 30 people or 1800 L.
The volume of people and furniture is 3600 L.
 $\% \text{ error} = (3600 \text{ L}/630,000 \text{ L})(100\%)$
 $= 0.57\%$.
- If die measures 1.55 cm on a side:
 $V = (1.55 \text{ cm})^3 = 3.72 \text{ cm}^3$
A die has 21 holes that are hemispheres with a radius of 0.20 cm.
Volume of hemisphere: $2/3\pi r^3$
 $V = 2/3\pi r^3 = 0.017 \text{ cm}^3$
Volume of 21 hemispheres: 0.36 cm^3
Volume of die:

$$V = 3.72 \text{ cm}^3 - 0.36 \text{ cm}^3$$

$$= 3.36 \text{ cm}^3$$

$$\text{Error} = 0.36 \text{ cm}^3$$

Percent error

$$= (0.36 \text{ cm}^3/3.36 \text{ cm}^3)(100\%) = 11\%$$

Note: The holes in some dice are cones. The volume of a cone is $1/3 \pi r^2 h$.

- Find weight in pounds and convert to kg.
Assume the density is about 1.00 kg/L.
 $V = \text{weight} \times 1 \text{ kg}/2.2 \text{ lb} \times 1 \text{ L}/1.00 \text{ kg}$

Section Review 4.1

Part A Completion

- | | |
|--------------|-------------------------------------|
| 1. atoms | 3. compounds |
| 2. different | 4. separated, joined, or rearranged |

Part B True-False

- | | | |
|-------|-------|-------|
| 5. NT | 6. ST | 7. AT |
|-------|-------|-------|

Part C Matching

- | | |
|------|-------|
| 8. c | 10. d |
| 9. a | 11. b |

Part D Questions and Problems

- simple whole-number ratios
- $\frac{1 \times 10^6 \text{ atoms}}{\text{cm}^3} \times \frac{100 \text{ cm}^3}{\text{m}} = 10^{10} \text{ atoms/m}$

Section Review 4.2

Part A Completion

- | | |
|--------------|----------------------|
| 1. subatomic | 5. nucleus |
| 2. negative | 6. Ernest Rutherford |
| 3. protons | 7. positive |
| 4. neutrons | 8. electrons |

Part B True-False

- | | |
|--------|--------|
| 9. NT | 11. NT |
| 10. AT | 12. AT |

Part C Matching

- | | | |
|-------|-------|-------|
| 13. c | 15. e | 17. b |
| 14. a | 16. d | |

Part D Questions and Problems

- protons, neutrons

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Judd E. Hollander



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Chemistry: The Central Science Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Darius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation Analysis of Kinetic Data E. S. Swinbourne, 2013-04-09 Data analysis is important from two points of view first it enables a large mass of information to be reduced to a reasonable compass and second it assists in the interpretation of experimental results against some framework of theory The purpose of this text is to provide a practical introduction to numerical methods of data analysis which have application in the field of experimental chemical kinetics Recognizing that kinetic data have many features in common with data derived from other sources I have considered it appropriate to discuss a selection of general methods of data analysis in the early chapters of the text It is the author's experience that an outline of these methods is not always easy to locate in summary form and that their usefulness is often not sufficiently appreciated Inclusion of these methods in the early chapters has been aimed at simplifying discussion in the later chapters which are more particularly concerned with kinetic systems By the provision of a number of worked examples and problems it is hoped that the reader will develop a feeling for the range of methods available and for their relative merits Throughout the text the mathematical treatment has been kept relatively simple lengthy proofs being avoided I have preferred to indicate the sense and usefulness of the various methods rather than to justify them on strict mathematical grounds *Sensors and Biosensors, MEMS Technologies and its Applications* Sergey Yurish, 2014-07-14 Sensors and Biosensors MEMS Technologies and its Applications Book Series Advances in Sensors Reviews Vol 2 18 chapters with sensor related state of the art reviews and descriptions of the latest achievements written by experts from academia and industry from 12 countries China India Iran Malaysia Poland Singapore Spain Taiwan Thailand UK Ukraine and USA This volume is divided into three main parts physical sensors biosensors nanoparticles MEMS technologies and applications With this unique combination of information in each volume the Advances in Sensors Reviews Book Series will be of value for scientists and engineers in industry and at universities to sensors developers distributors and users Like the 1st volume of this Book Series the 2nd volume also has been organized by topics of high interest *Mechanisms of Reactions of Metal Complexes in Solution* Debabrata Banerjee, M K Bharty, 2022-09-21 Reaction Mechanisms of Metal Complexes in Solution

provides a comprehensive overview of an often overlooked research area. Despite its importance and recent reshaping of the field, many inorganic chemists have lost an appreciation for the significance of stability constants and the thermodynamic aspects of complex formation. Ideal for newcomers and established researchers in the field, this book is a complete treatment of the area covering advanced topics with relevance to biomedical applications, extraction metallurgy, food chemistry, and a wealth of other industrial processes and research areas. The book will be of particular interest to postgraduates with an interest in coordination chemistry, catalysis, supramolecular chemistry, metallobiology, and related aspects of biochemistry.

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