

**Chapter 17 REVIEW**

1. Given a solution of ammonium chloride. What additional reagent or reagents are needed to prepare a buffer from the ammonium chloride solution?

Since  $\text{NH}_4\text{Cl}$  is a salt of a weak base, the weak base  $\text{NH}_3$  is needed.

- a) Explain how **this** buffer solution resists a change in pH when (show the reaction that occurs with explanation of how it buffers the solution):
- Moderate amounts of strong acid are added.



- Moderate amounts of strong base are added.



2. Describe the solubility of  $\text{CaCO}_3$  in each of the following solutions compared to its solubility in pure water. Show all applicable reactions. Give a brief explanation:

- a) In 0.10 M  $\text{NaCl}$       $\text{NaCl} \rightarrow \text{Na}^+ + \text{Cl}^-$  Neither of these ions are a common ion so there will be no effect on the solubility of  $\text{CaCO}_3$ .
- b) In 0.10 M  $\text{Na}_2\text{CO}_3$       $\text{Na}_2\text{CO}_3 \rightarrow 2\text{Na}^+ + \text{CO}_3^{2-}$  since this salt has carbonate which is a common ion it will affect the solubility of  $\text{CaCO}_3$ .      $\text{CaCO}_3(\text{s}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + \text{CO}_3^{2-}(\text{aq})$  so when the concentration of carbonate is increased it will shift the reaction to the left and decrease the solubility.
- c) In 0.10 M  $\text{HCl}$       $\text{HCl} \rightarrow \text{H}^+ + \text{Cl}^-$   $\text{H}^+$  will react with  $\text{CO}_3^{2-}$  to make carbonic acid which immediately breaks down into water and carbon dioxide.      $2\text{H}^+ + \text{CO}_3^{2-} \rightarrow \text{H}_2\text{O} + \text{CO}_2$  since the carbonate ion is reacted with the  $\text{H}^+$  it reduces the concentration of carbonate in the solution and causes the reaction to shift to the right increasing the solubility.

# Pearson Chemistry Ch 17 Key Terms

**IM Harris**



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**and Concepts of Organic Chemistry** Milton Orchin,Roger S. Macomber,Allan R. Pinhas,R. Marshall Wilson,2005-07-08 This book is a basic reference providing concise accurate definitions of the key terms and concepts of organic chemistry Not simply a listing of organic compounds structures and nomenclatures the book is organized into topical chapters in which related terms and concepts appear in close proximity to one another giving context to the information and helping to make fine distinctions more understandable Areas covered include bonding symmetry stereochemistry types of organic compounds reactions mechanisms spectroscopy and photochemistry *Extraction Techniques for Environmental Analysis* John R.

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*Conceptual Physical Science* Paul G. Hewitt,John Suchocki,Leslie A. Hewitt,2004 Conceptual Physical Science Third Edition takes learning physical science to a new level by combining Hewitt s leading conceptual approach and friendly writing style in a new edition that provides stronger integration of the sciences more quantitative coverage and a wealth of new media resources to help professors in class and students out of class The book s consistent high quality coverage includes five new chapters on chemistry astronomy and earth science for an even more balanced approach to physical science New Looking Forward and Looking Back boxes connect themes and concepts throughout the book helping students see the big picture More computational coverage eg Figuring Physical Science in chapter calculation allows students to practice the quantitative skills they need to master the concepts of physical science and be able to apply their knowledge Looking Forward and Looking Back boxes in every chapter connect themes and concepts throughout the book helping students see the big picture of physical science Powerful media package includes a comprehensive suite of award winning interactive online tutorials that offer students 24 7 help A media gri      **Chemistry 'O' Level** Rex M. Heyworth,2007

Academic Vocabulary Amy E. Olsen,2006-07 Interactive aesthetic workbook that uses high interest readings to show

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**Science** Richard T. Wright,Nancy M. Ostiguy,2005 **Chemistry** Catherine E. Housecroft,Edwin C. Constable,2010 This text integrates the three major branches of chemistry with the aim of enabling students to tackle more easily the problems within the subject and to apply chemistry to real life situations *Indian Journal of Chemistry. Section A. Inorganic, Physical, Theoretical, and Analytical* ,1986

**Measurement Across the Sciences** Luca Mari,Mark Wilson,Andrew Maul,2023-02-25 This open access book proposes a conceptual framework for understanding measurement across a broad range of scientific fields and areas of application such as physics engineering education and psychology It addresses contemporary issues and controversies within measurement in light of the framework including operationalism definitional uncertainty and the relations between measurement and computation and describes how the framework operating as a shared concept system supports understanding measurement s work in different domains using examples in the physical and human sciences This revised and expanded second edition features a new analysis of the analogies and the differences between the error uncertainty related approach adopted in physical measurement and the validity related approach adopted in psychosocial measurement In addition it provides a better analysis and presentation of measurement scales in particular about their relations with quantity units and introduces the measurand identification definition as a part of the Hexagon Framework along with new examples from the physical and psychosocial sciences Researchers and academics across a wide range of disciplines including biological physical social and behavioral scientists as well as specialists in measurement and philosophy appreciate the work s fresh and provocative approach to the field at a time when sound measurements of complex scientific systems are increasingly essential to solving critical global problems

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