

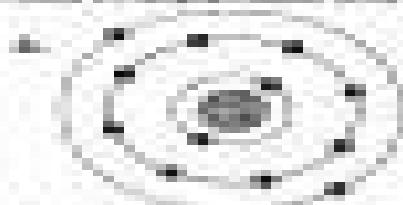
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IPS Unit 9 – Chemical Bonds

Types of Bonds Worksheet

Directions: Study the diagrams below, then give a brief description of each type of bond.



3. Explain the bond demonstrated in:

- Two atoms sharing one pair of electrons _____
- Two atoms sharing two pairs of electrons _____
- What is the charge of atom A? _____
- What is the charge of atom B? _____
- What is the net charge of the compound formed? _____

6. What type of bond will form?
ionic bond

7. Explain why a covalent bond is stronger than ionic bonds and why it is stronger:

Because there are two electrons shared between the atoms in a covalent bond, whereas in an ionic bond, the electrons are transferred from one atom to another.

Directions: Complete the table comparing the properties of ionic and covalent compounds.

| Compounds | Ionic compounds | Covalent compounds |
|-------------------------------------|----------------------|----------------------------|
| Electrical conductivity in solution | conducts electricity | do not conduct electricity |
| Boiling points | high | low |
| State of matter at room temperature | solid | liquid or gas |

Section 2 Reinforcement Types Of Bonds Answer Key

Liying Dong



Section 2 Reinforcement Types Of Bonds Answer Key:

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