

## 1. Identify the attractive forces associated with...

## a. Ionic Bonds:

Attractive electrostatic forces between negative charged ion and adjacent positively charged ion

## b. Covalent Bonds:

Electrostatic forces between negatively charged electrons and adjacent positively charged nucleus

## 2. Identify two similarities between ionic and covalent bonds

- Both involve valence electron clouds
- Both involve electrostatic attractions between oppositely charged species
- The formation of bond begins with the valence electrons of two atoms experiencing the attractive force of adjacent positive nuclei
- Both Ionic and Covalent bonds are strong

## 3. Identify two difference between ionic and covalent bonds

- Ionic Bonds involve the transfer of valence electrons forming ions whereas covalent bonds involve the sharing of valence electrons with no ion formation
- Ionic bonds form only between metals and non-metals whereas covalent bonds usually (but not always) form between two non-metals
- Ionic bonding does not result in molecule formation whereas covalent bonding usually does
- The attractive forces associated with covalent bonds are the electrostatic forces between negatively charged electrons and adjacent positively charged nuclei

## 4. Complete the following table by writing formulas of the compounds formed from the pairs of elements.

Elements	Compound Formula	$\Delta EN$ Value	Nature of Bonds Present
(a) rubidium and oxygen	$Rb_2O$	2.7	ionic
(b) strontium and bromine	$SrBr_2$	1.8	ionic
(c) carbon and sulphur	$CS_2$	0	covalent
(d) silicon and chlorine	$SiCl_4$	1.2	polar covalent

5. Magnesium is a metal and sulphur is a non-metal. Compare the  $\Delta EN$  value for these elements in  $MgS$  to  $\Delta EN$  value for the non-metal hydrogen and oxygen in water ( $H_2O$ ). Which of the two compounds possesses a greater amount of ionic character in its bond?

$\Delta EN$  for  $MgS$  = 1.3

$\Delta EN$  for  $H_2O$  = 1.4

We see that the bonds in water actually possess slightly more ionic character than those in magnesium sulphide even though the former compound contains two non-metals and the latter compound contains a metal and a non-metal

## Section 2 Types Of Bonds Worksheet Answer Key

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Christie, 2020-11-30 A family travels south in hopes of straightening out their misbehaving son only to experience violence and racism that threaten their safety The worksheets are laid out in a way that makes sense for students to follow along with the novel Put events from the novel in the order they occur at Kenny s school Predict whether Kenny will be able to mend his friendship with Rufus Explain why Kenny and his mother react a certain way to Byron s hairstyle choice Put yourselves into the author s shoes and speculate why Kenny s near drowning experience was put into a certain perspective Design a sign to post at Collier s Landing to warn people that it is not safe to swim there Identify elements of literature from the novel with a story elements graphic organizer Aligned to your State Standards and written to Bloom s Taxonomy additional crossword word search comprehension quiz and answer key are also included About the Novel The Watsons Go to Birmingham 1963 follows Kenny and his family as they experience firsthand the violence and racism engulfing the South in the 1960s Kenny is a fourth grader living with his family in Flint Michigan The winters are cold and the family must huddle together for warmth At school Kenny experiences bullying from his classmates only saved by his brother the biggest bully of all who picks on

Kenny at home The two look after their younger sister as she starts to learn right from wrong in Sunday school As Kenny's brother's behavior starts to get out of control their mom and dad decide to send him down to Birmingham Alabama to live with his grandmother over the summer Hoping she'll straighten him out the family sets off in their Brown Bomber on the long drive down south When they arrive they experience fear violence and racism that culminates in a tragedy that shakes them to their core From this experience Kenny must overcome his insecurities and deal with real world issues at a very young age

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