

## Chemistry 2<sup>nd</sup> Semester Final Review KEY

\*This is meant as a guide. It does NOT contain every single thing you need to know for the final exam. Review notes, benchmark review sheets, homeworks, chapter study guides, etc.\*

Bring your book to school by Monday for turn in (seniors should bring book to bookstore by themselves ASAP)

Gas Laws (Ch. 13):

Subtopics: atmospheric pressure, Dalton's law of partial pressures, combined gas law, Ideal gas law, gas stoichiometry (22.4 L/mol for a gas at STP)

Practice questions:

1. If a gas occupies 3.8 L at a pressure of 2.71 atm, what would the pressure be if the volume changes to 1.47 L? (Temp. and amt. of gas constant)

$$V_1 = 3.8 \text{ L}; P_1 = 2.71 \text{ atm}$$

$$V_2 = 1.47 \text{ L}; P_2 = ? \text{ atm}$$

Since all units are standard (L, atm), we do not need to convert the units before proceeding.

$$(P_1 V_1) / T_1 = (P_2 V_2) / T_2$$

T goes away (not mentioned in problem)

$$P_1 V_1 = P_2 V_2$$

Isolate for P2:  $P_2 = (P_1 V_1) / V_2$

$$P_2 = (2.71 \text{ atm} \times 3.8 \text{ L}) / (1.47 \text{ L}) = 7.0 \text{ atm} \text{ (L cancelled and it's 2 sig figs)}$$

2. A gas has a volume of 1.49 L at a temperature of 34.75 degrees Celsius. What would the volume be at 78.41 degrees Celsius? (pressure & amt. of gas constant)

$$V_1 = 1.49 \text{ L}, T_1 = 34.75 \text{ degrees Celsius}$$

$$V_2 = ? \text{ L}, T_2 = 78.41 \text{ degrees Celsius}$$

We need to convert from Celsius to Kelvins before proceeding.

$$T_1 = 34.75 + 273 = 307.75 \text{ K}$$

$$T_2 = 78.41 + 273 = 351.41 \text{ K}$$

$$(P_1 V_1) / T_1 = (P_2 V_2) / T_2$$

Get rid of P

$$V_1 / T_1 = V_2 / T_2$$

Isolate V2:  $V_2 = (V_1 / T_1) T_2$

$$V_2 = (1.49 \text{ L} / 307.75 \text{ K}) (351.41 \text{ K}) = 1.70 \text{ L}$$

3. What volume is occupied by 8.47 g of hydrogen gas at 84.7 degrees Celsius and 1.04 atm?

The combined gas law won't work here. You only have one condition and nothing changes.

$PV = nRT$  will work. ( $n$ = moles and  $R = 0.08026 \text{ L atm/K mol}$ )

$P = 1.04 \text{ atm}$ ,  $V = ? \text{ L}$ ,  $n =$  OOPS we need to convert from grams to moles!,  $T = 84.7 \text{ degrees Celsius} + 273 = 357.7 \text{ K}$

$n$ : Hydrogen gas is diatomic!

$$8.47 \text{ g H}_2 \times (1 \text{ mol H}_2 / 2.02 \text{ g H}_2) = 4.193069307 \text{ mol H}_2$$

$$\text{Isolate V: } V = (nRT) / P = (4.193069307 \text{ mol H}_2 \times 0.08026 \text{ L atm/K mol} \times 357.7 \text{ K}) / 1.04 \text{ atm} = 118 \text{ L}$$

4. What volume is occupied by 56.75 g of oxygen gas at STP?

# Second Semester Chemistry Finals Study Guide Answers

**Edward C. Gruber**

## **Second Semester Chemistry Finals Study Guide Answers:**

**Biology** Edward C. Gruber,1963    [61 Sample Question Papers for ICSE Class 10 Semester II Exam 2022 Oswal - Gurukul,2022-01-30](#)    *Blended Learning Designs in STEM Higher Education* Christopher N. Allan,Chris Campbell,Julie Crough,2019-04-09 This book offers a set of learning principles to support the design of rich learning experiences in Science Technology Engineering and Mathematics STEM higher education including detailed evaluations and discussions for a variety of science subjects Further it presents a professional learning framework that can be used to support the implementation of blended learning technologies to increase buy in from academic staff to support grass roots initiatives to develop a sense of community and to sustain change The principles developed here will help readers to think about blended learning from a learner s perspective put learning first and develop activities that will help learners achieve better learning outcomes In addition the book addresses how to design rich evidence based blended learning experiences that support learning It demonstrates a range of learning principles in practice with step by step instructions and includes templates supporting material instructions and other resources to help teachers embed and adapt designs in their own subject Readers will be equipped with an expanded toolkit of resources designs ideas and activities that can be directly applied in a variety of subject areas

*MCAT 2015: What the Test Change Means for You Now* Kaplan,2014-08-05 Big changes are coming to the MCAT in 2015 and Kaplan is here to help you prepare for them With four brand new sections 80% more questions and the addition of new science content including biochemistry psychology and sociology the 2015 MCAT will be a completely different test In order to be prepared you need to understand the exam and start planning for it now and this guide is the first step MCAT 2015 What the Test Change Means for You Now is your complete guide to the new exam with outlines of both old and new subject areas a short form practice test to help you get ready and advice on choosing and prepping for the MCAT that s right for you

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designed to assess the examinee's problem solving critical thinking and knowledge of science concepts and principles prerequisite to the study of medicine Scores are reported in Physical Sciences Verbal Reasoning and Biological Sciences Study guide covers AAMC Association of American Medical Colleges content Physical Sciences Biological Sciences Verbal Reasoning Mathematics Concepts The Cell Chromosomes Reproduction Implantation Microorganisms Biochemistry Human Physiology The Heart The Lymphatic System GI Tract Musculoskeletal System Kidney Hormones Nerves Skin Genetics Populations and Evolution Elements Hund's Rule and Radiation The Periodic Table Covalent Bonds Molecular Shapes General Chemistry Kinetic Molecular Theory Phase Change Solutions Oxidation Numbers Entropy Acids and Bases Galvanic and Electrolytic Cells Carbon Stereochemistry Alkanes and Alkenes Hydrogen Bonding Alcohols Phenols Aldehydes and Ketones in Carboxylic Acids Ether Ammonia Amino Acids Carbohydrates Spectroscopy Separation and Distillation Vectors and Simple Motion Forces Circular and Projectile Motion Statics Center of Gravity Work and Energy Power and Momentum Stress and Strain Elasticity and Density Hydrostatic Pressure Fluids in Motion Electricity and Magnetism Plates Capacitors Voltage Batteries Resistors Magnetic Fields Waves and Periodic Motion Sound Waves Doppler Effect Simple Harmonic Motion Optics Mirrors 205 pages

Learning While Black Janice E. Hale, 2001-12-04 In Learning While Black Janice Hale argues that educators must look beyond the clichés of urban poverty and teacher training to explain the failures of public education with regard to black students Why Hale asks simply are black students not being educated as well as white students Hale goes beyond finger pointing to search for solutions Closing the achievement gap of African American children she writes does not involve better teacher training or more parental involvement The solution lies in the classroom in the nature of the interaction between the teacher and the child And the key she argues is the instructional vision and leadership provided by principals To meet the needs of diverse learners the school must become the heart and soul of a broad effort the coordinator of tutoring and support services provided by churches service clubs fraternal organizations parents and concerned citizens Calling for the creation of the beloved community envisioned by Dr Martin Luther King Jr Hale outlines strategies for redefining the school as the Family and the broader community as the Village in which each child is too precious to be left behind In this book I am calling for the school to improve traditional instructional practices and create culturally salient instruction that connects African American children to academic achievement The instruction should be so delightful that the children love coming to school and find learning to be fun and exciting Janice Hale

*Study Guide & Solutions Manual to Accompany Organic Chemistry* Mary H. Bailey, 1998 **Resources in Education**, 1994 **Me n Mine-Science-Term-2** Saraswati Experts, A text book on science General Chemistry Workbook Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student

will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material     *Book of Abstracts American Chemical Society. Meeting,American Chemical Society,1973*     **Guide to Class 9 All India SAINIK School Entrance Exam (AISSEE) with 2 Practice Sets - 2nd Edition** Disha Experts,2022-05-07 The Guide to AISSEE Class 9 All India SAINIK School Entrance Exam with 2 Practice Sets 2nd Edition provides complete Preparatory Material Solved Papers Practice Sets The book covers the 5 sections of the exam Intelligence Mathematics English General Science and Social Studies The book provides exhaustive theory with examples followed by exercise in each chapter It also provides past 6 year Questions papers 2016 22 included chapter wise There are 100 chapters in all The book provides 2600 questions for practice Answers to most of the questions are provided The book also provides 2 Practice Sets on the latest pattern of the exam at the end of the book     **Field Assistant and Claims Assistant** Arco Publishing Company,1949     *Guide to AISSEE Class 9 All India SAINIK School Entrance Exam with 2 Practice Sets & Previous Year Questions - 4th Edition* Disha Experts, The 4th Edition of Guide to AISSEE Class 9 All India SAINIK School Entrance Exam with 2 Practice Sets Previous Year Questions 4th Edition provides complete Preparatory Material Solved Questions Practice Sets The book covers the 5 sections of the exam Intelligence Mathematics English General Science and Social Studies The book provides exhaustive theory with examples followed by exercise in each chapter It also provides past 9 year Questions papers 2016 24 included chapter wise There are 100 chapters in all The book provides 2900 questions for practice Answers to most of the questions are provided The book also provides 2 Practice Sets on the latest pattern of the exam at the end of the book     **Health Inspector, Department of Health ...** Arco Publishing Company,1949     **Implementation and Critical Assessment of the Flipped Classroom Experience** Scheg, Abigail G.,2015-01-31 In the past decade traditional classroom teaching models have been transformed in order to better promote active learning and learner engagement Implementation and Critical Assessment of the Flipped Classroom Experience seeks to capture the momentum of non traditional teaching methods and provide a necessary resource for individuals who are interested in taking advantage of this pedagogical endeavor Using narrative explanations and foundation materials provided by experienced instructors this premier reference work presents the benefits and challenges of flipped methodology implementation in today s classroom to educators and educational administrators across all disciplines and levels     Annual Catalogue United States Air Force Academy,1984

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