

Your Full Name/Lab Section Kylie HenkelSynthesis of Aspirin
Pre-Lab Assignment

Read the lab handout before attempting these questions. SHOW WORK for ALL calculations. Pay attention to significant figures.

1. A. (1) What functional group is undergoing a transformation in this reaction?

Alcohol

- B. (1) What functional group is the functional group in 1A being transformed into (in the product)?

Ester

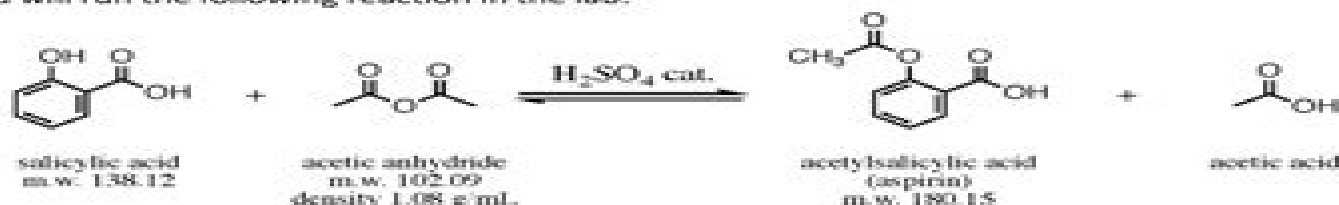
2. (2) What **physical property** must you include in your table of reagents for acetylsalicylic acid (Aspirin)? Why do you need to include it?

We need to include melting point in the table because it is relevant in Part B of the experiment - analysis

3. (1) What specific solvent system (what solvents and in what ratio) are you using for the TLC analysis of your product?

3 : 2 Hex : EtOAc

4. You will run the following reaction in the lab:



- A. (1) Assume you use 2.00 mL of acetic anhydride in the reaction. Use the density of acetic anhydride to determine the mass of acetic anhydride used. Show work.

$$D = \frac{m}{V}$$
$$D = 1.08 \text{ g/mL}$$
$$V = 2.00 \text{ mL}$$
$$\xrightarrow{2.00 \text{ mL}} 1.08 \text{ g/mL} = \frac{m}{2.00 \text{ mL}} \sqrt{2.00 \text{ mL}}$$
$$m = 2.16 \text{ g}$$

Techniques Of Organic Chemistry Pre Lab Answers

Didier Musso



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Table of Contents Techniques Of Organic Chemistry Pre Lab Answers

1. Understanding the eBook Techniques Of Organic Chemistry Pre Lab Answers
 - The Rise of Digital Reading Techniques Of Organic Chemistry Pre Lab Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Techniques Of Organic Chemistry Pre Lab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Techniques Of Organic Chemistry Pre Lab Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Techniques Of Organic Chemistry Pre Lab Answers
 - Personalized Recommendations
 - Techniques Of Organic Chemistry Pre Lab Answers User Reviews and Ratings
 - Techniques Of Organic Chemistry Pre Lab Answers and Bestseller Lists
5. Accessing Techniques Of Organic Chemistry Pre Lab Answers Free and Paid eBooks
 - Techniques Of Organic Chemistry Pre Lab Answers Public Domain eBooks
 - Techniques Of Organic Chemistry Pre Lab Answers eBook Subscription Services
 - Techniques Of Organic Chemistry Pre Lab Answers Budget-Friendly Options
6. Navigating Techniques Of Organic Chemistry Pre Lab Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Techniques Of Organic Chemistry Pre Lab Answers Compatibility with Devices
 - Techniques Of Organic Chemistry Pre Lab Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Techniques Of Organic Chemistry Pre Lab Answers
 - Highlighting and Note-Taking Techniques Of Organic Chemistry Pre Lab Answers
 - Interactive Elements Techniques Of Organic Chemistry Pre Lab Answers
 8. Staying Engaged with Techniques Of Organic Chemistry Pre Lab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Techniques Of Organic Chemistry Pre Lab Answers
 9. Balancing eBooks and Physical Books Techniques Of Organic Chemistry Pre Lab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Techniques Of Organic Chemistry Pre Lab Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Techniques Of Organic Chemistry Pre Lab Answers
 - Setting Reading Goals Techniques Of Organic Chemistry Pre Lab Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Techniques Of Organic Chemistry Pre Lab Answers
 - Fact-Checking eBook Content of Techniques Of Organic Chemistry Pre Lab Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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