

Physical Science – 2nd Semester – Final Exam Study Guide – (2012–2013)

1. Physical Science Basics

- What tool(s) would you want to use to find the:
 - Mass of an object? triple beam balance Basic SI Unit? kg
 - Volume of an object? ruler or water displacement method Basic SI Unit? cm³ or ml
 - Length of an object? Ruler Basic SI Unit? meter
 - Density of an object? graduated cylinder Basic SI Unit? ml
- What is the formula for calculating density? mass/volume
 - What is the density of a cube that is 25 g and has a volume of 3 cm³? 8g/cm³
 - Compare the density of solid H₂O to liquid H₂O, and explain what observable effect this has.
solid H₂O less dense than liquid H₂O

2. Chemistry Basics

- Element (def.) substance made up of one kind of atom
- Atom (def.) basic building block of matter
- Compound (def.) a molecule that contains atoms or molecules of two or more different elements
- Molecule (def.) a combination of two or more of the same or different atoms that are bonded together

3. Model of the Atomic Evolution of the Atomic Theory

Atoms are so small that many scientists throughout history have created models to describe them.

Why have atomic theories changed over time? Better technology came around.

Scientist	Year	Major Discovery/ Contribution	Name of atomic model
John Dalton	1808	Each element is made up of small, indivisible particles called atoms. Different elements have atoms of different masses.	Billiard Ball Model
J.J. Thomson	1897	Negatively charged electron	Plum Pudding Model
Niels Bohr	1913	Electrons move <u>fixed orbits</u> around the atom's nucleus in specific layers, or <u>electron shells</u> . Atoms absorb or give off energy when the electrons move from one shell (<u>energy level</u>) to another.	Bohr Model
Ernest Rutherford	1909	1909 /Based on idea that positive reflects positive/ atom mostly empty space; electrons orbit randomly around positively charged nucleus	Planetary Model

4. Periodic Table

- The current periodic table is arranged by increasing atomic number.
- Reading the Periodic Table
 - An element's properties can be predicted from its location in the periodic table.
 - The atomic number represents the number of protons.
 - The atomic mass represents the number of protons and neutrons.

Phycical Science 2nd Semester Final Exam

Emilie Sanchez



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Peterson's Graduate Programs in the Physical Sciences 2011 Peterson's, 2011-05-01 Peterson's Graduate Programs in the Physical Sciences contains a wealth of information on colleges and universities that offer graduate work in Astronomy and Astrophysics Chemistry Geosciences Marine Sciences and Oceanography Meteorology and Atmospheric Sciences and Physics The institutions listed include those in the United States Canada and abroad that are accredited by U S accrediting bodies Up to date information collected through Peterson's Annual Survey of Graduate and Professional Institutions provides valuable information on degree offerings professional accreditation jointly offered degrees part time and evening weekend programs postbaccalaureate distance degrees faculty students degree requirements entrance requirements expenses financial support faculty research and unit head and application contact information As an added bonus readers will find a helpful See Close Up link to in depth program descriptions written by some of these institutions These Close Ups offer detailed information about the physical sciences program faculty members and their research and links to the program or department's Web site In addition there are valuable articles on financial assistance and support at the graduate level and the graduate admissions process with special advice for international and minority students Another article discusses important facts about accreditation and provides a current list of accrediting agencies

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Annual Catalog United States Air Force Academy, 1983 [Annual Catalog - United States Air Force Academy](#) United States Air Force Academy, 1971 **United States Air Force Academy** United States Air Force Academy, **Graduate Programs in the Physical Sciences, Mathematics, Agricultural Sciences, the Environment & Natural Resources**

2011 (Grad 4) Peterson's, 2011-05-01 Peterson's Graduate Programs in the Physical Sciences Mathematics Agricultural Sciences the Environment Natural Resources contains a wealth of information on colleges and universities that offer graduate work in these exciting fields The institutions listed include those in the United States and Canada as well international institutions that are accredited by U S accrediting bodies Up to date information collected through Peterson's Annual Survey of Graduate and Professional Institutions provides valuable information on degree offerings professional accreditation jointly offered degrees part time and evening weekend programs postbaccalaureate distance degrees faculty students degree requirements entrance requirements expenses financial support faculty research and unit head and application contact information Readers will find helpful links to in depth descriptions that offer additional detailed information about a specific program or department faculty members and their research and much more In addition there are valuable articles on financial assistance the graduate admissions process advice for international and minority students and facts about accreditation with a current list of accrediting agencies Research in Education ,1974 *Collected Papers of Carl Wieman* Carl E. Wieman, 2008 Carl Wieman's contributions have had a major impact on defining the field of atomic physics as it exists today His ground breaking research has included precision laser spectroscopy using lasers and atoms to provide important table top tests of theories of elementary particle physics the development of techniques to cool and trap atoms using laser light particularly in inventing much simpler less expensive ways to do this the understanding of how atoms interact with one another and light at ultracold temperatures and the creation of the first Bose Einstein condensation in a dilute gas and the study of the properties of this condensate In recent years he has also turned his attention to physics education and new methods and research in that area This indispensable volume presents his collected papers with annotations from the author tracing his fascinating research path and providing valuable insight about the significance of the works *Guide to the Evaluation of Educational Experiences in the Armed Services* American Council on Education, 1978 2004 Physics Education Research Conference Jeffrey Marx, Paula Heron, Scott Franklin, 2005-09-29 The 2004 Physics Education Research PER Conference brought together researchers in how we teach physics and how it is learned Student understanding of concepts the efficacy of different pedagogical techniques and the importance of student attitudes toward physics and knowledge were all discussed These Proceedings capture an important snapshot of the PER community containing an incredibly broad collection of research papers of work in progress **Official Gazette** Philippines, 2005 *American Universities and Colleges* ,1987 *Congressional Record* United States. Congress, 1958 The Congressional Record is the official record of the proceedings and debates of the United States Congress It is published daily when Congress is in session The Congressional Record began publication in 1873 Debates for sessions prior to 1873 are recorded in The Debates and Proceedings in the Congress of the United States 1789 1824 the Register of Debates in Congress 1824 1837 and the Congressional Globe 1833 1873 *Resources in Education* ,1989 **Directory of Distance**

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