

Name\_\_\_\_\_

Period\_\_\_\_\_

**Unit 8: Stoichiometry Test Review***Indicate whether the statement is true or false. (1-11)*

- \_\_\_\_\_ 1. *Stoichiometry* is the study of the relationship between the amount of reactants used and the amount of products made in a chemical reaction.
- \_\_\_\_\_ 2. In a balanced chemical equation, the total number of moles of all the reactants is equal to the total number of moles of all the products.
- \_\_\_\_\_ 3. In a chemical reaction represented by the general equation  $A + B \rightarrow C + D$ , there are four distinct mole ratios that can be written.
- \_\_\_\_\_ 4. The *mole ratio* is a comparison of how many moles of one substance are required to participate in a chemical reaction with another substance, based on the balanced chemical equation.
- \_\_\_\_\_ 5. In a chemical reaction, the reactant with the largest molar mass is the limiting reagent.
- \_\_\_\_\_ 6. The *excess reagent* refers to the additional amount of a reactant in a chemical reaction that must be added in order for the reaction to proceed.
- \_\_\_\_\_ 7. The stoichiometric relationship between any two substances in a reaction depends on the mole ratio between those substances.
- \_\_\_\_\_ 8. The *limiting reagent* limits the amount of product formed in a chemical reaction.
- \_\_\_\_\_ 9. In a balanced chemical equation, the total number mass of the reactants is equal to the total mass of the products.
- \_\_\_\_\_ 10. The percent yield is the maximum amount of product that can be produced from a given amount of reactant.
- \_\_\_\_\_ 11. If 80.0 g of sodium reacts with 72.0 g of iron (III) oxide, sodium acts as the limiting reactant.

- \_\_\_\_\_ 12. Based on the mole ratios of the substances in a chemical reaction shown, determine the correct equation for the chemical reaction?

Substances	Mole ratio
A:B	3:2
A:C	3:1
B:C	2:1

- \_\_\_\_\_ 13. Calculate the mole ratio of all the substances represented in this figure?

## Unit 8 Test Stoichiometry

**Mark C. M. van Loosdrecht, Per  
Halkjaer Nielsen, C. M. Lopez-  
Vazquez, Damir Brdjanovic**



## Unit 8 Test Stoichiometry:

Elementary Principles of Chemical Processes Richard M. Felder, Ronald W. Rousseau, Lisa G. Bullard, 2016 Elementary Principles of Chemical Processes 4th Edition Student International Version prepares students to formulate and solve material and energy balances in chemical process systems and lays the foundation for subsequent courses in chemical engineering The text provides a realistic informative and positive introduction to the practice of chemical engineering MCAT General Chemistry Review 2024-2025 Kaplan Test Prep, 2023-07-04 Kaplan's MCAT General Chemistry Review 2024 2025 offers an expert study plan detailed subject review and hundreds of online and in book practice questions all authored by the experts behind the MCAT prep course that has helped more people get into medical school than all other major courses combined

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**Improving Student Comprehension of Stoichiometric Concepts** Connie Lynn Bannick Kemner,2007

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