Periodic Trends: Ionization Energy Worksheet

- 1. What is the ionization energy of an element?
- 2. Predict the following trends you expect to see in ionization energy:

 a) As you move across a period? Why?
 - b) As you move down a group? Why?
- 3. What factors influence the ease of removing an electron from an element?
- 4. Choose an orbital where an electron would experience the highest Z_{eff} effective nuclear charge (least shielded), and the highest first ionization energy. Circle the correct answer.

Na (3s) Mg (3s) Al (3p) P (3p) S (3p)

5. Match the following electron configuration with the appropriate ionization energy.

a) 1s2 2s2 2p6 3s2 3p6 4s2 3d10 4p6 5s1

0 1356 kJ/mol

b) 1s2 2s2 2p6 3s2 3p6 4s2

ii) 595 kJ/mol

c) 1s² 2s² 2p⁶ 3s² 3p⁶ 4s² 3d¹⁰ 4p⁶

iii) 409 kJ/mol

Which one of the following has the largest first ionization energy? Circle the correct answer.

Na K Li Cs

7. The first ionization energies values (kJ/mol) for sequential elements in period 3 are shown below. Assign elements to these values based on the periodic trends.

740 578 786 1012 999

 Choose the element with the greatest first ionization energy. Circle the correct answer.

a) Carbon or aluminum

b) Calcium or strontium

c) Helium or lithium

d) Chlorine or argon

Skill Practice 17 Ionization Answers

Marco Cascella

Skill Practice 17 Ionization Answers:

Chemistry All-in-One For Dummies (+ Chapter Quizzes Online) Christopher R. Hren, John T. Moore, Peter J. Mikulecky, 2022-11-23 Everything you need to crush chemistry with confidence Chemistry All in One For Dummies arms you with all the no nonsense how to content you ll need to pass your chemistry class with flying colors You ll find tons of practical examples and practice problems and you ll get access to an online quiz for every chapter Reinforce the concepts you learn in the classroom and beef up your understanding of all the chemistry topics covered in the standard curriculum Prepping for the AP Chemistry exam Dummies has your back with plenty of review before test day With clear definitions concise explanations and plenty of helpful information on everything from matter and molecules to moles and measurements Chemistry All in One For Dummies is a one stop resource for chem students of all valences Review all the topics covered in a full year high school chemistry course or one semester of college chemistry Understand atoms molecules and the periodic table of elements Master chemical equations solutions and states of matter Complete practice problems and end of chapter quizzes online Chemistry All In One For Dummies is perfect for students who need help with coursework or want to cram extra hard to ace that chem test

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