IIT-JEE (ADVANCED)

CHAPTER TEST-1

CHEMISTRY

Top	pic : Some	basic conce	pts of chemi-	stry (STOIC	HIOMETRY)

[SOLUTIONS]

$$\frac{14.2}{71} \text{ gm} = 0.2 \text{ mole}$$

$$\frac{\text{moles of KO}_2}{2} = \frac{\text{moles of H}_2\text{O}_2}{1}$$

$$\frac{0.2}{2} = \frac{\text{moles of H}_2\text{O}_2}{1}$$

$$\frac{\text{actual}}{\text{theoretical}} \times 100 = \frac{1.7}{3.4} \times 100$$

$$= 50\%$$
moles of H2O2 = 1

0

40%
$$x\%$$
 $60 - x\%$

$$\frac{40}{12} \qquad \frac{x}{1} \qquad \frac{60 - x}{16}$$
Given: $x : \frac{60 - x}{16} = 2 : 1$

$$\frac{x}{\frac{60-x}{16}} = \frac{2}{1}$$

(3)

$$16x = 120 - 2x$$

$$x = \frac{120}{18} = \frac{20}{3}$$
C H

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Michael Brown

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