

Chemistry A - Unit 8 Sample Work/Study Guide

Instructions

- Download & save to your computer in word document format OR make a copy in google docs.
- Complete this as you work along in your lessons. This is to be turned in at the end of the unit
- Can be used as your notes for the test.
- Do **NOT** look up any answers online - this will be a violation of our school's Honor Code Policy! Use the LESSONS to answer these questions. Teachers review BRAINLY, QUIZLET, other online sources **REGULARLY** and know the answers posted.

★ Unit 8 MINI Lessons -watch these recordings to help with this unit ★

Lesson 2 - Electrical Bonding Investigation

1. A substance's phase (solid, liquid, gas) depends on the amount of kinetic energy of the molecules
2. Electrostatic forces - forces that hold the molecules together in a solid or liquid. These come from the different charges in the molecules
3. As heat energy is added to a solid substance, the molecules gain kinetic energy and move around more. If they gain enough kinetic energy they can overcome the electrostatic forces and form a liquid.
4. Vapor Pressure - caused from molecules near the surface gaining enough energy to turn to gas. More gas molecules causes more pressure.
5. Atoms - molecules held together by electrostatic attraction between surrounding molecules

Lesson 3 - Electrical Bonding Investigation Plan

1. What are 5 Bulk properties of a substance?

5 Bulk properties of a substance include melting point, boiling point, freezing point, vapor pressure, and surface tension.

2. Define Electrostatic force - Electrostatic forces hold molecules, or atoms, together in a solid or liquid. It is the attraction, or repulsion between charged particles.
3. How can we determine which substance has stronger or weaker bonds?

We can determine which substance has a stronger or weaker bond by looking at its freezing point, and boiling point. A higher boiling point usually means the substance has strong electrical forces between molecules. A higher freezing point usually means the substance has stronger electrostatic forces.

Unit 8 Study Guide Chemistry Answers

Mary Wuerth



Unit 8 Study Guide Chemistry Answers:

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