

## Acids and bases review sheet Answer key

1. List the differences between an acid and a base. Your description should include such things as whether both are electrolytes, which contains a hydrogen ion and which includes a hydroxide ion, and you should also include two examples of an acid and two examples of a base that are present in your household.

Acid	base
Conducts electricity	conducts electricity
Contains hydrogen ion	contains hydroxide ion
Proton donor	proton acceptor
Turns litmus paper red	turns litmus paper blue
React with most metals	releases hydroxide ion
Releases hydronium	Ex: Windex,
Ex: lemon	bleach
Vinegar	soap
Battery acid (sulfuric acid)	

2. Listed below are some of the properties of acids and bases. Fill in the blanks with the appropriate word, **acids** or **bases**:

- A**   produce hydrogen ions ( $H^+$ ) in solution
- B**   have a bitter taste
- B**   produce hydroxide ions ( $OH^-$ ) ions
- A**   react with bases to form salts
- A**   have a sour taste
- B**   have a slippery, soapy feel
- B**   react with acids to form salts
- A**   react with many metals

3. Write an equation for the dissociation of the following **acids** in water; follow example

- $HClO_4 \rightarrow H^+ + ClO_4^-$
- $H_2SO_4 \rightarrow 2H^+ + SO_4^{2-}$
- $HC_2H_3O_2 \rightarrow H^+ + C_2H_3O_2^-$
- $H_2S \rightarrow 2H^+ + S^{2-}$
- $HCl \rightarrow H^+ + Cl^-$
- $HNO_3 \rightarrow H^+ + NO_3^-$

4. Write an equation for the dissociation (ionization) of the following **bases** in water: (follow example)

- $Mg(OH)_2 \rightarrow Mg^{2+} + 2OH^-$
- $NaOH \rightarrow Na^+ + OH^-$
- $Ba(OH)_2 \rightarrow Ba^{2+} + 2(OH^-)$
- $KOH \rightarrow K^+ + OH^-$
- $Mg(OH)_2 \rightarrow Mg^{2+} + 2(OH^-)$
- $LiOH \rightarrow Li^+ + OH^-$

5) A. What type of reaction is a **neutralization** reaction? **Double replacement**

B. Write the net equation for neutralization:



# Properties Of Acids And Bases Answers

**Mr. Rohit Manglik**



## Properties Of Acids And Bases Answers:

*Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual* Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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## **Table of Contents Properties Of Acids And Bases Answers**

1. Understanding the eBook Properties Of Acids And Bases Answers
  - The Rise of Digital Reading Properties Of Acids And Bases Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Properties Of Acids And Bases Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Properties Of Acids And Bases Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Properties Of Acids And Bases Answers
  - Personalized Recommendations
  - Properties Of Acids And Bases Answers User Reviews and Ratings
  - Properties Of Acids And Bases Answers and Bestseller Lists
5. Accessing Properties Of Acids And Bases Answers Free and Paid eBooks
  - Properties Of Acids And Bases Answers Public Domain eBooks
  - Properties Of Acids And Bases Answers eBook Subscription Services
  - Properties Of Acids And Bases Answers Budget-Friendly Options



6. Navigating Properties Of Acids And Bases Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Properties Of Acids And Bases Answers Compatibility with Devices
  - Properties Of Acids And Bases Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Properties Of Acids And Bases Answers
  - Highlighting and Note-Taking Properties Of Acids And Bases Answers
  - Interactive Elements Properties Of Acids And Bases Answers
8. Staying Engaged with Properties Of Acids And Bases Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Properties Of Acids And Bases Answers
9. Balancing eBooks and Physical Books Properties Of Acids And Bases Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Properties Of Acids And Bases Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Properties Of Acids And Bases Answers
  - Setting Reading Goals Properties Of Acids And Bases Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Properties Of Acids And Bases Answers
  - Fact-Checking eBook Content of Properties Of Acids And Bases Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

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