

WORD EQUATIONS

Name Key

Write the word equations below as chemical equations and balance.

1. zinc + lead (II) nitrate yield zinc nitrate + lead



2. aluminum bromide + chlorine yield aluminum chloride + bromine



3. sodium phosphate + calcium chloride yield calcium phosphate + sodium chloride



4. potassium chlorate when heated yields potassium chloride + oxygen gas



- * 5. aluminum + hydrochloric acid yield aluminum chloride + hydrogen gas



- * 6. calcium hydroxide + phosphoric acid yield calcium phosphate + water



7. copper + sulfuric acid yield copper (II) sulfate + water + sulfur dioxide



8. hydrogen + nitrogen monoxide yield water + nitrogen



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Coordinating Procedural and Conceptual Knowledge to Make Sense of Word Equations: Understanding the Complexity of a 'simple' Chemical Task at the Learner's Resolution Keith Stephen Taber, Pat Bricheno, 2009 Abstract This paper discusses the conceptual demands of an apparently straightforward task set to secondary level students completing chemical word equations with a single omitted term Chemical equations are of considerable importance in chemistry and school students are expected to learn to be able to write and interpret them However it is recognized that many students find them challenging The present paper explores students accounts of their attempts to identify the missing terms to illuminate why working with chemical word equations is so challenging from the learner's perspective 300 secondary age students responded to a 5 item exercise based on chemicals and types of reactions commonly met at school level For each item they were asked to identify the missing term in a word equation and explain their answers This provided a database containing more than a thousand student accounts of their rationales Analysis of the data led to the identification of seven main classes of

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