# Ch 13 Study Guide (Chemical Bonds)

1. Fill in the table below with the correct values:

Element Name	lon Symbol	# Protons	# Electrons	Charge	Ion Type
Fluorine	F.	9	10	-1	Anion
lodine	I-	53	54	-1	Anien
Sulfar	52-	16	18	-2	Anien
Potassium	K*	19	18	+1	Cation
Calcium	Ca <sup>2+</sup>	20	18	+2	Carlon
Bromine	Be	35	36	-1	Anien
Streetliam	Sr <sup>2+</sup>	38	36	+2	Cation
Oxygen	O2-	8	10	- 2	Anion
Magnesium	Mag+	12	10	+2	Cation
Aluminum	413+	13	10	+3	Cation
Selenium	Sea-	34	36	-2	Anion
Lithium	Li*	3	2	+1	Carrion
Rubidium	Rb*	37	36	+1	Cation
Chlorine	CR-	17	18	-1	Anion

2. An ionic bond is formed between what two types of atoms?

Metal & nonmetal

3. A covalent bond is formed between what two types of atoms?

Nonmetal & Nonmetal

4. Draw the bond(s) that will form between Mg and O.

Mg . \_ => Mg :0:

5. What is the formula for the compound formed in question #4?

Mao

# **Worksheet 13 Chemical Bonding Answers**

Frank Schaffer Publications, Joan DiStasio

# **Worksheet 13 Chemical Bonding Answers:**

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