

1. Use Model 1 to show which atoms are present in each type of molecule by listing the symbol for each atom included. Carbohydrate has been done for you.

a. Carbohydrate— C, H, O

c. Amino acid— N, C, H, O, R, S

b. Lipid— C, H, O

d. Nucleic acid— N, C, H, O, P

2. Which type of molecule includes an example with a long-chain carbon backbone?

Lipids.

3. In the molecule referred to in the previous question, what is the dominant element attached to the carbon backbone?

Hydrogen (carbon is attached to hydrogen)

4. The fatty acid chain of the lipids is often referred to as a hydrocarbon chain. Discuss with your group why the chain is given this name and write a one-sentence definition for a hydrocarbon.

Fatty acid chains of the lipids is referred to as a hydrocarbon chain because carbon is constantly attached to hydrogen.

5. Which molecule has a central carbon atom with four different components around it?

Proteins.

amino acids
are made up
of nitrogen
carbon

6. Which molecule has a sugar, nitrogenous base, and phosphate group?

nucleic acids

7. Discuss with your group members some similarities among all four types of molecules. List as many as you can.

- carbon
- hydrogen
- oxygen

8. What is the chemical formula of the first carbohydrate molecule shown?



9. What three structural groups shown do all amino acids have in common?

They all consist of carbon, hydrogen, nitrogen, and oxygen (R side chain)

10. There are 20 naturally-occurring amino acids, and each one only varies in the structure of the R side chain. Two amino acids are shown in Model 1. What are the R side chains in each?



Read This!

During chemical reactions, the bonds in molecules are continually broken and reformed. To break a bond, energy must be absorbed. When bonds are formed, energy is released. If more energy is released than absorbed during a chemical change, the process can be used as a source of energy. A general rule for processes such as respiration is the more carbon atoms there are in a molecule, the more energy that molecule can provide to the organism when it is used as food.

Properties Of Biological Molecules Pogil Answers

**Charles R. Cantor, Paul Reinhard
Schimmel**



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Water and Biological Macromolecules Eric Westhof,1993 **An Introduction to the Structure of Biological Molecules** John Michael Barry,E. M. Barry,1969 Biophysical Properties of Biological Macromolecules Mats Eriksson,1995 Biological Molecules David Shackleton,1997 **Studies on the Conformations and Properties of Biomolecules** B. Robson,University of Manchester. Department of Chemistry,1983 **Biological Molecules** Chris A. Smith,Edward J. Wood,1978 Origin of Chirality in the Molecules of Life Albert Guijarro,Miguel Yus,2008-11-04 This book provides an interdisciplinary review of one of the great unsolved mysteries that has fascinated scientists for over 150 years the origin of chirality in biomolecules It was Pasteur who first initiated the search for a deterministic theory to explain the handedness of biomolecules His theory that a dissymmetric force was involved was correct in essence but he never saw the fruits of his labour Current thinking tells us that asymmetry in the universe has its origins in the forces that unfolded after the Big Bang and more specifically the weak force Being left handed the weak force imprinted its signature on the evolving Universe However at the molecular level the weak force does not provide a straightforward explanation of biomolecular homochirality In fact it is yet to be proved beyond doubt that a causal link exists at all Many alternative theories have been put forward some of them resting on solid ground but all lacking definitive experimental evidence to back them up Some postulate that the handedness of molecules in the biosphere arose by chance but this is hard to test Others rely on discovering life on similar planets and making comparisons with Earth Alternative theories have emerged from a range of backgrounds including geology biology chemistry physics and astronomy Current advances in fields as diverse as space exploration prebiotic chemistry and high energy physics may help to provide an answer Important pieces of information will come from observations at the two frontiers of science outer space and the subatomic world Observation of distant planets galaxies and even actual sampling of celestial objects from beyond the solar system are projects currently underway At the other end of the spectrum there are experiments that study the elemental properties of matter such as symmetry and interactions with the fundamental forces All these efforts will render their fruits soon This volume unifies all the theories of the origin of biomolecular homochirality together in one source The various chapters focus on chance mechanisms physical forces such as the weak interaction fluid dynamics amplification of chirality the organic contents of meteorites and comets and finally the physical view of an intrinsically asymmetric universe This complete interdisciplinary review of an intriguing subject condenses a large and disparate range of contributions from journals in almost every scientific field The various theories have been organized interrelated and explained in a unified way One of the book's strengths is its extensive use of graphic material to aid understanding the many subjects covered It is fundamental comprehensive and structured to be

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Biomolecules: Analysis and Properties BIOTOL,1992-09-01

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Structure and Stability of Biological Macromolecules Serge N. Timasheff,Gerald D. Fasman,1969

Biomolecules Serafín Fraga,Ester Fraga,1971

Giant Molecules: Here, There, And Everywhere (2nd Edition) Alexander Y Grosberg,Alexei R Khokhlov,2010-09-24

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