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 Section 22-33-7179 Team —
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Pre-Lab Study Questions | 17

1. In making pickles, a cucumber is placed in a concentrated salt solution. Explain what happens to the cucumber when it is left in the salt solution for some time.

The salt solution penetrates the cucumber's outer layer causing the water content to move outside of its semi-permeable membrane & the salt solution moves in.

2. Why is it important that cell membranes are semipermeable?

To allow small particles & water to move in & out of the cell membrane.

via osmosis; causes shrinkage

3. What are the differences between solutions, colloids, and suspensions?

- **Solutions:** homogeneous mixture of two or more components.
 - **Colloids:** smaller particles than suspensions / intermediate in size (vs) the others
 - **Suspensions:** Requires filtration to separate from solvent / settles out.
4. Indicate the compartment (A or B) that will increase in volume for each of the following pairs of solutions separated by a semipermeable membrane:

Solution A	Solution B
a. 10% (m/v) starch	4% (m/v) starch
b. 2% (m/v) albumin	5% (m/v) albumin
c. 8% (m/v) sucrose	0.8% (m/v) sucrose

Solution (A)
Solution (B)
Solution (A)

5. Indicate whether a red blood cell will undergo hemolysis, crenation, or no change in each of the following:

- a. 10% (m/v) NaCl
- b. 2% (m/v) glucose
- c. H_2O
- d. 5% (m/v) glucose

crenation
hemolysis
hemolysis
no change

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