${\sf PDF}$

Substitute symbols and formulas for words:

- Predict the products.
- 31 Include abbreviations to denote physical state.
- 4) Balance each equation.
- When heated, solid silver oxide decomposes into its elements.

 Solutions of potassium chloride and silver nitrate are maked to produce a potassium compound, which dissolves, and a silver compound, which is insoluble.

 Write the equation for the synthesis reaction between magnesium metal and axygen gos.

4. A piece of copper metal is placed into a solution of zinc nitrate.

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5. Write the equation for the combustion of propose gas.

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