

Worksheet - Osmosis & Tonicity

READ FIRST: In each diagram below, a "cell" with a semipermeable membrane has been placed in a beaker containing substances that are dissolved in water. The membrane is **permeable** to water & iodine. It is **not permeable** to glucose, sodium (Na⁺), or starch. *Please remember that unlike sugar's solution, an indicator for starch! Therefore, it will turn from yellow-brown to blue-black in the presence of starch.* If not otherwise indicated, you may assume for each problem that the remainder of the solution is water.

Beaker 1

- What is the % of water inside the cell? _____
- What is the % of water outside the cell? _____
- Will osmosis occur? _____
- If so, in what direction will osmosis occur? _____
- Will glucose diffuse? _____
- Will the cell shrink or swell? _____
- How do you know? _____
- This diagram shows the cell is a(n) (circle one) hypotonic / hypertonic / isotonic solution.



Beaker 2

- What is the % of water inside the cell? _____
- What is the % of water outside the cell? _____
- Will osmosis occur? _____
- If so, in what direction will osmosis occur? _____
- Will glucose diffuse? _____
- Will the cell shrink or swell? _____
- How do you know? _____
- This diagram shows the cell is a(n) (circle one) hypotonic / hypertonic / isotonic solution.



Beaker 3

- What is the % of water inside the cell? _____
- What is the % of water outside the cell? _____
- Will there be a net change in these concentrations? _____
- Will osmosis occur? _____ Why?
- Will starch diffuse? _____ Will glucose diffuse? _____
- If iodine were placed in the beaker, what would you see immediately?
- What would you see after several hours? Why?
- This diagram shows the cell is a(n) (circle one) hypotonic / hypertonic / isotonic solution.



Worksheet Osmosis And Tonicity Answers

D Keegan

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