

Answer each of the following questions using the equation provided. BE SURE TO BALANCE EACH EQUATION BEFORE SOLVING ANY PROBLEMS. SHOW ALL WORK.



- a. 2 moles of NO will react with _____ mole(s) of O₂ to produce _____ mole(s) of NO₂.

b. $7 \text{ moles NO}_2 = 3.6 \text{ moles O}_2 \times \frac{\text{moles NO}_2}{\text{moles O}_2} =$

- c. How many moles of NO must react to form 4.67 moles of NO₂?



- a. 20 moles of NH₃ are needed to produce _____ moles of H₂O.

- b. How many moles of N₂ will be produced if 3.5 moles of O₂ react?

Worksheet Word Equations Chemistry A Study Of Matter

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Chemistry Brady,1996-05-01

Chemistry, the Study of Matter and Its Changes, James E. Brady, John R. Holum Michael Kenney,1996

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