



Chapter 1

Temperature, Thermal Energy, and Heat

Temperature: a measure of how hot or cold an object is, based on the average kinetic energy of the particles that make up the object. It is related to the average kinetic energy of the particles that make up the object.

How _____

Explain the concept _____

How _____

Explain the concept _____

How _____

Explain _____

How _____

Explain the concept _____

Explain the concept _____

How _____

Explain the concept _____

How _____

1. Temperature is a measure of the average kinetic energy of the particles that make up the object.
2. The higher the temperature, the higher the average kinetic energy of the particles that make up the object.
3. Temperature is a measure of the average kinetic energy of the particles that make up the object.
4. The higher the temperature, the higher the average kinetic energy of the particles that make up the object.
5. The higher the temperature, the higher the average kinetic energy of the particles that make up the object.
6. The higher the temperature, the higher the average kinetic energy of the particles that make up the object.
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18. The higher the temperature, the higher the average kinetic energy of the particles that make up the object.
19. The higher the temperature, the higher the average kinetic energy of the particles that make up the object.
20. The higher the temperature, the higher the average kinetic energy of the particles that make up the object.

Thermal Energy: the energy associated with the random motion of the particles that make up the object.

How _____

- a. Thermal energy is the energy associated with the random motion of the particles that make up the object.

How _____

- a. Thermal energy is the energy associated with the random motion of the particles that make up the object.

How _____

- a. Thermal energy is the energy associated with the random motion of the particles that make up the object.

How _____

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