Section 5.2 The Modern Periodic Table (pages 130–138)

This section explains the organization of the modern periodic table and discusses the general properties of nectals, nonnectals, and nectaliside.

Reading Strategy (page 199)

Previewing Before you read, complete the table by writing two questions about the periodic table on pages 132–135. As you read, write answers to your questions. For more information on this reading studges, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.

Owestions About the Periodic Tuble	
Discussion	Answer
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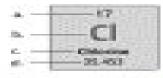
Efter possible greations Why are there has immitating systems to the optional FWhy is parted 7 incomplate?

The Periodic Law (pages 101-100)

- Is the following sentence true or false? In the modern periodic table, elements are amonged by increasing number of protons.
- Explain why the number of elements per period varies. The number of elements per period varies because the number of available orbitals increases from energy level to energy level.
- Proporties of observents repeat in a productable way velses atomic numbers are used to arrange elements into groups. This pattern of repeating properties is called the

Atomic Mass tone 1040

 Label the four types of information supplied for chlorine in the diagram.



- the Alberta surebar
- Education (special)

C. Element corne

of Attaces comm.

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1 which of the following statements accurately describes animals a all animals are multicellular all are heterotrophic and all lack cell walls b all animals are multicellular some are heterotrophic and some lack cell walls c some animals are multicellular all are heterotrophic and all lack cell walls d

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